Section II—Milk and Food

MILK SURVEILLANCE

Although milk is only one of the many sources of dietary intake of radionuclides, it is the single food item most often used as an indicator of the population's intake of radionuclides from the environment. This is because fresh milk is consumed by a large segment of the United States population and contains most of the radionuclides occurring in the environment which have been identified as biologically important. In addition, milk is produced and consumed on a regular basis, is convenient to handle, is easily analyzed, and samples which are representative of milk consumption in any area can be readily obtained.

1. Pasteurized Milk Network, November 1964

Division of Radiological Health and Division of Environmental Engineering and Food Protection, Public Health Service

The Public Health Service pasteurized milk surveillance program had its origin in a raw milk monitoring network (1) established by the Service in 1957. One of the primary objectives of the raw milk network was the development of methods for milk collection and radiochemical analysis suitable for larger scale programs.

Experience derived from this earlier network led to the activation of a pasteurized milk sampling program with stations selected to provide nationwide surveillance of milk production and consumption areas. The present network, which consists of 63 stations, has at least one station in every State, the Canal Zone, and Puerto Rico.

Sampling procedure

Through the cooperation of State and local milk sanitation authorities, samples are routinely collected at each station. The method specifies that each station's sample be composited of subsamples from each milk processing plant in proportion to the plant's average sales in the community served. At most stations the sample represents from 80 to 100 percent of the milk processed. Prior to September 15, 1961, the composite sample was taken from one day's sales per month and was as representative of the community's supply as could be achieved under practical conditions. Beginning with the resumption of nuclear weapons testing in the atmosphere in September 1961, and continuing through January 1963, samples were collected twice a week at nearly all stations and daily for short periods at selected stations. Since then, the sampling frequency has been reduced to once a week.

Samples are preserved with formaldehyde and sent to the PHS Southwestern (SWRHL), Southeastern (SERHL), or Northeastern Radiological Health Laboratories (NERHL) for analysis. Gamma analyses for iodine-131 are made within 3 to 6 days after sample collection, and any results exceeding 100 pc/liter are immediately telephoned to State health officials for possible public health action. Analytical results are normally available 6 to 7 weeks after monthly samples are received by the laboratories; publication in RHD follows 3 to 4 months after the monthly samples are composited for analyses.

Iodine-131, cesium-137, and barium-140 concentrations are determined by gamma scintillation spectroscopy.1 After the weekly samples are gamma scanned, samples from two consecutive weeks are composited and analyzed radiochemically for strontium-89 and strontium-90. There is an inherent statistical variation associated with all measurements of radionuclide concentrations. With the low radionuclide levels which are usually found in milk and other environmental samples, this variation on a percentage basis is relatively high. The variation depends upon such factors as the method of chemical analysis, the sample counting rate and counting time, interferences from other radionuclides, and the background count. For milk samples, counting times of 50 minutes per gamma spectroscopy and 30 to 50 minutes for beta determinations are used. Table 1 shows the approximate total analytical error (including counting error) associated with radionuclide concentrations in milk. These errors were determined by comparing results of a large number of replicate analyses. Table 1 gives the 95 percent confidence limits between which the true concentrations of the selected radionuclides might be expected in the analyses. The minimum detectable concentration is defined as the measured concentration at which the two-standard-deviation analytical error is equal to the measurement. Accordingly, the minimum detectable concentrations in units of pc/liter are Sr⁸⁹, 5; Sr⁹⁰, 2; Cs¹³⁷, 10; Ba¹⁴⁰, 10; and I131, 10. At these levels and below, the counting error comprises nearly all of the analytical error.

TABLE 1.—ANALYTICAL ERRORS ASSOCIATED WITH ESTIMATED CONCENTRATIONS FOR SELECTED RADIONUCLIDES IN MILK

Nuclide	Estimated concentration (pc/liter)	Error a (pc/liter)	Estimated concentration (pc/liter)	Error a (percent of concentra- tion)
Iodine-131	0 to 100	±10	100 or greater	±10
Barium-140	0 to 100	±10	100 or greater	±10
Cesium-137	0 to 100	±10	100 or greater	±10
Strontium-89	0 to 50	± 5	50 or greater	±10
Strontium-90	0 to 20	± 2	20 or greater	±10

^a Two standard deviations (2σ).

Calcium analyses at SERHL are done by an ion exchange and permanganate titration method, while at NERHL and SWRHL an ethylenediaminetetraacetic acid (EDTA) method is used. Stable potassium concentrations are estimated from the potassium-40 concentrations² determined from the gamma spectrum.

Data presentation

Table 2 presents summaries of the analyses for November 1964 (November 1-28). Radionuclide values reported by a laboratory as being below the minimum detectable concentration have been averaged by using one-half the minimum detectable value. The averaging procedure was modified for iodine-131 and barium-140 in October 1963 when nondetectable concentrations of these radionuclides were considered zero. A similar procedure is used for the network average.

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Figures 1 and 2 are isogram maps showing the estimated strontium-90 and cesium-137 concentrations in milk over the entire country. The value printed beside each station is the monthly average concentration for that station. The isograms were developed by arbitrary interpolation between values for the individual stations. Additional modifications to the isograms are made according to available information on milksheds.

In order to develop the distribution of the network's stations *versus* radionuclide concentrations in milk, tables 3 and 4 have been prepared from the monthly averages shown in table 2, and November 1963 data has been included for comparison.

The average monthly strontium-90 concentrations in pasteurized milk from selected cities in the sampling program are presented in figure 3. The data regarding reported nuclear detonations given in this figure are presented for information only. The underground nuclear detonations during 1963 and 1964 are not intended to imply a cause and effect relationship with the rising strontium levels during this period.

Each graph shows the strontium-90 concentrations in milk from one city in each of the four U.S. Bureau of Census regions. This method of selection permits graphic presentation of data for each city in the network three times a

¹ Southeastern Radiological Health Laboratory employs a radiochemical procedure for barium-140 analysis.

² The conversion factor is 1.18 x 10⁻⁹ g K/pc K⁴⁰.

TABLE 2.—RADIOACTIVITY IN PASTEURIZED MILK, NOVEMBER 1964

8	Sampling locations		ium iter)	Stronti (pc/l		Stronti (pc/l		Cesius (pc/l	m-137 iter)		e-131 liter)		m-140 liter)
		Third quarter	Avg. for month	Third quarter	Avg. for month	Third quarter	Avg. for month	Third quarter	Avg. for month	Third quarter	Avg. for month	Third quarter	Avg. for month
la: laska: riz: rk: alif:	Montgomery Palmer Phoenix Little Rock Sacramento San Francisco	1.17 1.17 1.15 1.15 1.26 1.22	1.20 1.27 1.20 1.18 1.31 1.23	<5 <5 <5 <5 <5 <5	<5 <5 <5 <5 <5 <5	21 16 3 38 6 7	20 19 4 33 6 4	65 105 25 80 30 35	60 80 20 60 30 35	0 0 0 0 0	0 0 0 0 0	0 0 0 0 0	0 0 0 0 0 0
. Z: olo: onn: el: . C:	Cristobal	1.11 1.25 1.10 1.15 1.11 1.19	1.14 1.24 1.16 1.19 1.14 1.19	<5 <5 <5 <5 <5 <5	<5 <5 <5 <5 <5 <5	5 20 19 20 19 17	5 16 14 16 17 14	50 100 110 85 65 255	50 60 90 70 40 210	0 0 0 0 0	0 0 0 10 0	0 0 0 0 0	000000000000000000000000000000000000000
a: lawaii: laho: l: nd:	Atlanta Honolulu Idaho Falls Chicago Indianapolis	1.17 1.19 1.24 1.14 1.15	1.19 1.20 1.24 1.20 1.22	<5 <5 <5 <5 <5	<5 5 <5 <5 <5	28 11 25 19 18	25 15 21 16 15	105 70 110 85 70	90 70 85 70 60	0 0 0 0	0 0 0 0 10	0 0 0 0	000000000000000000000000000000000000000
owa: ans: y: a: laine:	Des Moines Wichita Louisville New Orleans Portland	1.17	1.23 1.27 1.19 1.23 1.23	<5 <5 <5 <5 <5	<5 <5 <5 15 <5	22 18 27 48 29	23 19 24 40 26	70 50 50 115 175	50 30 45 70 155	0 0 0 0	0 0 0 0 10	0 0 0 0	000
Id: Mass: Mich:	Baltimore	1.13	1.17 1.17 1.19 1.23	<5 <5 <5 <5	<5 <5 <5 <5	21 31 16 19	19 23 16 20	70 180 90 100	50 130 75 90	0 0 0	10	0 0 0	
Minn: Miss: Mo:	Minneapolis Jackson Kansas City St. Louis Helena	1.21 1.19 1.20	1.24 1.23 1.22 1.24 1.26	<5 <5 <5 <5 <5	5 5 <5 <5 <5	27 37 26 21 20	23 30 21 18 15	115 85 55 55 115	75 60 40 40 85	0 0 0 0	0 0	0 0 0 0	
Nebr: Nev: N. H: N. J: N. Mex:	Omaha Las Vegas Manchester Trenton Albuquerque	1.19 1.17 1.12	1.13 1.25 1.18 1.14 1.24	<5 <5 <5 <5 <5	<5 <5 <5 <5 <5	24 6 30 18 9	18 10 22 17 9	70 50 205 85 45	40 55 165 70 45	0 0 0 0	10 10	0 0 0 0	1
N. Y: N. C: N. Dak:	Buffalo New York Syracuse Charlotte Minot	1.09 1.12 1.17	1.14 1.20 1.14 1.21 1.23	<5 <5 <5 <5 10	<5 <5 <5 <5 <5	18 25 15 38 49	16 22 16 30 35	105 120 80 100 145	90 95 90 65 100	000000000000000000000000000000000000000	10	0	1
Ohio: Okla: Ore: Pa:	Cincinnati Cleveland Oklahoma City Portland Philadelphia Pittsburgh	1.12 1.12 1.24 1.11	1.18 1.18 1.30 1.20	<5 <5 <5	<5 5 10 <5 <5 <5	19 19 19 30 18 29	18 18 23 16	90 50 165 80	60 70 45 85 60 80	000000000000000000000000000000000000000	0 0 0	0000	
P. R: R. I: S. C: S. Dak: Tenn:	San Juan Providence Charleston Rapid City Chattanooga Memphis	1.13 1.15 1.19 1.20	1.18 1.18 1.10 1.20	<5 <5 <5 <5	<5 <5 10	43	18 28 32 31	140 110 125 90	95 90 110 65	0	0 0	0	1
Tex: Utah: Vt: Va:	Austin Dallas Salt Lake City Burlington Norfolk	1.14 1.27 1.13	1.18 1.34 1.20	<5 <5 <5	<5 <5 <5	17 25 25	15	45 165 140	30 65 110				
Wash: W. Va: Wis: Wyo:	Seattle Spokane Charleston Milwaukee Laramie	1.22 1.13 1.19	1.16	<5 <5 <5	<5 <5 <5	26 24 14	17	130 50 95	95 40 80				
Network	k average	1.17	1.21	<5	< 5	22.3	19.1	94	72		0 <10) (<1

year. The last column in table 2 shows the most recent issue in which a graph of the strontium-90 concentration was given for each station. A tabulation of the network monthly

maximum, minimum, and average radionuclide concentrations in milk was given for March 1960-March 1964 in the July 1964 issue of RHD (2).

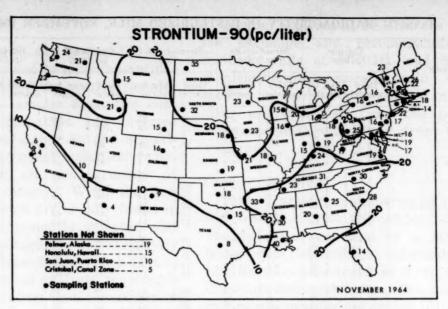


FIGURE 1.—STRONTIUM-90 CONCENTRATIONS IN PASTEURIZED MILK, NOVEMBER 1964

Table 3.—RANGES OF STATION MONTHLY AVERAGES FOR STRONTIUM-90, JUNE-NOVEMBER 1964 . AND NOVEMBER 1963

	Number of stations in range								
Range (pc/liter)		1964							
1.1	June	July	Aug.	Sept.	Oct.	Nov.	Nov.		
Under 10	4	6 6	7	8	6	6			
10-19	6		23 23	8 27 20	37	31			
20-29	23	29			13	19	3		
30-39	17	14	6	6	7	6	13		
40-49	10	4	4	2	0	1	- 1		
50-59	10	1	0	0	0	0			
60-69	0	2	O	0	0	l ŏ			
70-79		Ō	o o	0	Ö	Ŏ			

Table 4.—RANGES OF STATION MONTHLY AVERAGES FOR CESIUM-137, JUNE-NOVEMBER 1964 AND NOVEMBER 1963

		Number of stations in range							
Range (pc/liter)	1964								
	June	July	Aug.	Sept.	Oct.	Nov.	Nov.		
Under 50	1 19	4 24	6 30	14 33	16 40	15 41	21		
100-149 150-199	19 24 13	21	19	13	6	4 2	21 22 10		
200-249 250-299	4 2	8 3 2	1	1 0	0	0	1		

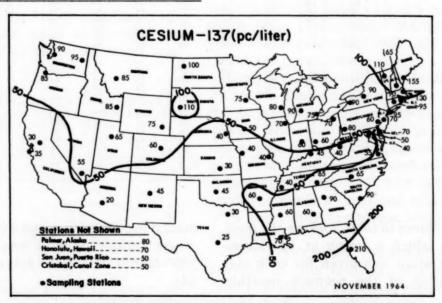


FIGURE 2.—CESIUM-137 CONCENTRATIONS IN PASTEURIZED MILK, NOVEMBER 1964

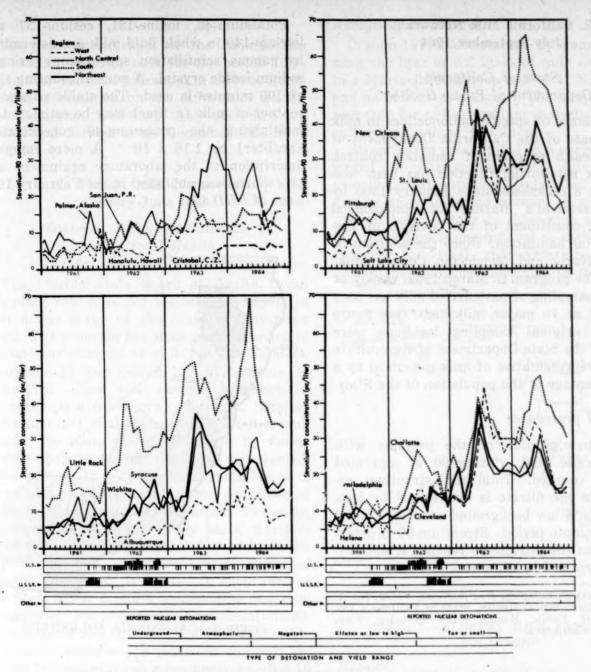


FIGURE 3.—STRONTIUM-90 IN PASTEURIZED MILK, 1961-NOVEMBER 1964

March 1965

Data

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2. California Milk Network3, July-September 1964 ·

State of California Department of Public Health

Surveillance of specific radionuclides in milk is one phase of the California Department of Public Health program of radiation control. This milk monitoring function has been conducted at 8 milksheds since January 1960 by the Department's Bureau of Radiological Health, a constituent of the Division of Environmental Sanitation. Since the addition of the Del Norte and Mendocino County milksheds to the program in March 1962, weekly or biweekly sampling of pasteurized milk has been conducted at 10 major milksheds (see figure 4). The original sampling locations were chosen by the State Department of Agriculture as being representative of milk consumed by a high percentage of the population of the State.

Analytical procedures

After precipitation of the proteins with trichloroacetic acid, yttrium-90 is separated and beta counted. Total radiostrontium remaining in the filtrate is determined by beta counting in a low background counter, usually for a 60-minute period. Strontium-90 is determined from the yttrium-90 results; strontium-89 is determined by difference.

Potassium-40, iodine-131, cesium-137 and barium-140 in whole fluid milk are determined by gamma scintillation spectroscopy using a sodium-iodide crystal. A normal counting time of 100 minutes is used. The stable potassium content of milk (g/liter) may be estimated by multiplying the potassium-40 concentration (pc/liter) by 1.18 x 10⁻³. A more complete description of the laboratory equipment and procedures was published in the February 1963 issue of RHD (3).



FIGURE 4.—CALIFORNIA MILKSHEDS

TABLE 5.—RADIONUCLIDES IN CALIFORNIA MILK, JULY-SEPTEMBER 1964

[Radioactivity concentrations in pc/liter]

Element an	d month	Del Norte	Fresno	Hum- boldt	Los Angeles	Mendo- cino	Sacra- mento	San Diego	Santa Clara	Shasta	Sonoma	Average
Calcium (g/liter)	July	1.25 1.33 1.30	1.20 1.10 1.20	1.24 1.14 1.28	1.20 1.00 0.96	1.15 1.20 1.20	1.18 1.10 1.20	1.10 1.10 1.20	1.20 1.10 1.20	1.05 1.10 1.20	1.20 1.20 1.20	1.1
Potassium-40	July	1,460 1,220 1,010	1,200 1,210 1,190	1,170 1,140 1,070	1,210 1,230 1,200	1,230 1,190 1,210	1,170 1,170 1,210	1,250 1,270 1,230	1,190 1,320 1,180	1,240 1,180 1,240	1,220 1,280 1,210	1,234 1,221 1,175
Strontium-89	July Aug Sept	14 b	b b	2 6	b b	2 b						
Strontium-90	July	54 56 32	3.9 4.0 4.2	14 11 13	3.8 6.2 3.7	6.4 7.4 6.0	11.0 4.5 5.7	4.3 3.3 3.1	4.9 3.9 6.5	12.1 7.3 5.6	6.1 4.0 5.2	12.0 10.8 8.5
Cesium-137	July Aug Sept	155 110 54	26 18 18	40 27 18	23 16 14	31 28 18	25 26 21	15 13 11	34 20 24	44 41 25	34 27 18	43 33 22

a No significant amounts of iodine-131 or barium-lanthanum-140 in samples for this period were found.
b No detectable activity.

³ Data from Radiological Health News, Vol. 3, No. 4, Bureau of Radiological Health, State of California Department of Public Health, 2151 Berkeley Way, Berkeley 4, California.

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1.18 1.14 1.19 34 21 75 2

12.0 10.8 8.5 43 33 22

Data

The monthly averages of the radionuclide and calcium data for milk for the period July-September 1964 are presented in table 5. Background information on the differences among the California milksheds was presented by Heslep and Cornish in the December 1963 issue of RHD (4).

3. Florida Milk Network, January-June 1964

Division of Radiological and Occupational Health Florida State Board of Health

The Florida State Board of Health began sampling raw milk for iodine-131 analysis in two major areas of the State in November 1962. The program has since been expanded to include the analysis of milk for strontium-89, strontium-90 and cesium-137 in addition to iodine-131. Raw milk samples are received from the six areas shown in figure 5. Samples for iodine-131 analysis are taken from a tank truck, the route of which passes by farms widely dispersed over the area represented. Where there is no route representative of a large portion of the area, samples are collected from selected farms and combined. Presently, the sampling is on a monthly basis. Samples were collected weekly when iodine-131 was detectable in milk. A regional State Board of Health Laboratory is located in each area of the State where samples representing raw milk production are collected. For strontium-89, strontium-90 and cesium-137 analysis, monthly composite samples are prepared by each regional laboratory from a randomly selected 10 percent of the farms which have their milk regularly sampled for bacteriological analyses by the county health departments. These composite samples are then sent to the State Radiological Health Laboratory in Orlando for analysis. Milk produced in the counties comprising each area is generally processed, marketed, and consumed in that area. These areas are characterized by differences in dairying practices related to the gradual transition from small farms—using locally grown feeds in the "West Florida" region-to larger farms using different types of grass and predominantly purchased feeds in the southern areas.

Analytical procedures

Cesium-137 is determined by gamma scanning one liter or 3.5 liters of milk as received in a Marinelli beaker on a 4" x 4" NaI crystal and multichannel analyzer. The sample is then prepared for radiostrontium analysis by evaporating and ashing one liter.

Strontium is precipitated as a carbonate and stored for yttrium ingrowth. After the ingrowth period, yttrium is separated with carrier as an oxalate and counted in a low-background beta counter to determine strontium-90. Strontium is precipitated as carbonate and counted for total radiostrontium activity. Strontium-89 is then calculated by difference.



FIGURE 5.—FLORIDA MILK SAMPLING LOCATIONS

Results

Table 6 presents the available monthly average strontium-89 and strontium-90 concentrations in Florida raw milk during January-April 1964. Table 7 presents cesium-137 and iodine-131 results for January-June 1964. Strontium-89, strontium-90, and cesium-137 results for 1963 appeared in the October 1964 issue of RHD (5). The results for 1963 appeared in the March 1964 issue of RHD (6). A comparison of these data indicate that during the early part of 1964 Florida milk had lower strontium-89 levels but higher strontium-90 levels than were observed in the preceding 9 months.

TABLE 6.—STRONTIUM-89 AND STRONTIUM-90 IN FLORIDA RAW MILK, JANUARY-APRIL 1964

[pc/liter]

Radionuclide and month	West Florida	North Florida	Northeast Florida	Central Florida	Tampa Bay	Southeast Florida	Average
Strontium-89 January February March April May June	<5 <5 <5 <5						
Average	<5	<5	<5	<5	<5	<5	<5
Average (Jan-Dec 1963)	32	32	28	30	26	31	30
Strontium-90			1			L Trees	1000
January February March April	17 10 13 15	16 11 14 17	15 10 11 14	10 11 12 10	13 10 15	12 10 15 14	13 11 13 14
Average	13.8	14.5	12.5	10.8	11.5	12.8	12.
Average (Jan-Dec 1963)	10.4	9.6	9.1	10.1	8.7	9.2	9.

TABLE 7.—CESIUM-137 AND IODINE-131 IN FLORIDA RAW MILK, JANUARY-JUNE 1964

[pc/liter]

Radionuclide and month	West Florida	North Florida	Northeast Florida	Central Florida	Tampa Bay	Southeast Florida	Average
Cesium-187 January February March April	191 192 232 213	170 203 175 276	159 196 181 278	211 244 309 294	176 198 234 250	201 200 234 -251	185 206 228 260
May June	305 256	284 240	287 182	363 347	264 261	264 246	295 255
Average	232	225	214	295	231	233	238
Average (Apr-Dec 1963)	180	183	228	227	202	218	206
Iodine-131 January February March April May	<2 <2 <2 <2 <2 <10	=	<2 <2 <2 <2 <2 <2 <10	<2 <2 <2 <2 <2 <2 <10	<2 <2 <2 <2 <2 <2 <10	<2 <2 <2 <2 <2 <2 <10	<2 <2 <2 <2 <2 <2 <10

4. Oregon Milk Network, June-September 1964

Division of Sanitation & Engineering Oregon State Board of Health

The Oregon State Board of Health conducts milk monitoring at eight major milk-producing centers throughout the State of Oregon, as shown in figure 6. Half-gallon samples of pasteurized packaged milk are collected statewide on a monthly basis by the Oregon Department of Agriculture and weekly in the Portland area by the city of Portland. The sampling frequency is accelerated to a weekly schedule at those locations having radionuclide concentrations in milk in excess of 100 pc/liter for iodine-131 or 500 pc/liter for cesium-137. The samples are forwarded to the Oregon State

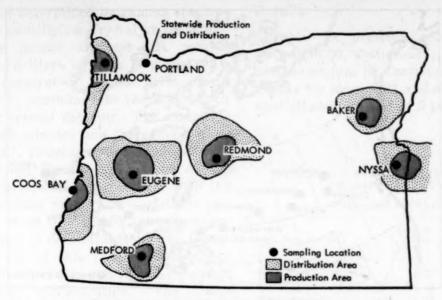


FIGURE 6.—OREGON PASTEURIZED MILK NETWORK SAMPLING LOCATIONS SHOWING PRODUCTION AND DISTRIBUTION AREAS

Board of Health environmental radiation laboratory for iodine-131, cesium-137, barium-140, and strontium-90 analyses. The gamma analyses are performed utilizing a 3" x 3" sodium iodide scintillation detector with a 512-channel gamma spectrometer. Samples are normally counted for 100 minutes. The strontium-90 concentrations reported by the Oregon State Board of Health were obtained by the trichloracetic acid analytical procedure (7) with a low background beta counter equipped with a 2inch detector used for counting. The minimum detectable concentrations for iodine-131, cesium-137, and barium-140 are 15 pc/liter. The minimum detectable concentration is defined to be that amount of activity which, in the same counting time, gives a count that differs from the background count by 3 times the standard deviation (σ) of the background

Table 8 presents the Oregon milk surveillance data for the period June through September 1964. The Portland composite sample represents contributions from nearly all milksheds in Oregon, plus some in southern Washington. Thus, it tends to represent a State average.

Table 8.—RADIONUCLIDE CONCENTRATIONS IN OREGON MILK, JUNE-SEPTEMBER 1964 •

[Average concentrations in pc/liter]

Sampling location and frequency	Nuclide	June	July	Aug.	Sept.
Baker	Strontium-90	22	29	28	100
Monthly	Cesium-137	115	105	135	
Coos Bay	Strontium-90	43	29	21	65
Monthly	Cesium-137	135	105	90	
Eugene	Strontium-90	14	12	18	75
Monthly	Cesium-137	125	110	95	
Medford	Strontium-90	15	22	13	55
Monthly	Cesium-137	125	145	90	
Nyssa	Strontium-90	25	31	18	60
Monthly	Cesium-137	125	130	90	
Portland composite	Strontium-90	42	28	• 27	* 20
Weekly	Cesium-137	223	197	143	96
Portland local producer	Strontium-90	24	33	36	90
Weekly	Cesium-137	168	160	134	
Redmond Monthly	Strontium-90 Cesium-137	100	20 120	20 80	65
Tillamook Monthly	Strontium-90 Cesium-137	340	29 190	37 160	180

Iodine-131 and barium-140 concentrations were below the lower limit of detection of 15 pc/liter.
 Dash indicates no analysis performed.
 Analyses performed by Public Health Service.

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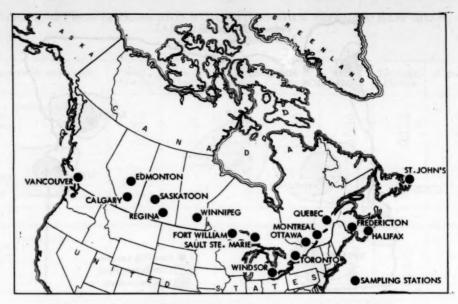


FIGURE 7.—CANADIAN MILK SAMPLING STATIONS

5. Canadian Milk Network⁴ November 1964

Radiation Protection Division,
Department of National Health and Welfare,
Ottawa, Canada

The Radiation Protection Division of the Department of National Health and Welfare began monitoring milk for strontium-90 in November 1955. At first, analyses were carried out on samples of powdered milk obtained from processing plants. However, since January 1963 liquid whole milk has been analyzed instead. With this change, more representative samples of milk consumed can be obtained, and in addition it is possible to choose milk sampling locations (see figure 7) in the same areas as the air and precipitation stations. At present the analyses include determinations of iodine-131, strontium-89, cesium-137, strontium-90 as well as stable potassium and calcium.

The milk samples are obtained through the cooperation of the Marketing Division of the Canadian Department of Agriculture. At each station samples are collected three times a week from selected dairies and are combined into weekly composites and forwarded to the radiochemical laboratory in Ottawa. The con-

tribution of each dairy to the composite sample is directly proportional to its volume of sales. In most cases a complete sample represents over 80 percent of the milk processed and distributed in the area. Several of the weekly samples are randomly selected and analyzed for iodine-131. The results of the spot checks for iodine-131 will not be reported unless there is evidence that the levels are rising. A monthly composite of the samples is analyzed for strontium-90, cesium-137, and stable potassium and calcium.

Analytical methods

Radiochemical methods are used for the analysis of iodine-131 (8). For the analysis of radiostrontium, carrier strontium is added to a one-liter sample of milk, and the milk is then placed in a tray lined with a polyethylene sheet and evaporated under infra-red lamps. The residue is ashed in a muffle furnace at 450 degrees C., dissolved in dilute nitric acid, and strontium separated by fuming nitric acid precipitation. The combined strontium-89 and strontium-90 are determined by counting in a low background beta counter. Strontium-90 is determined separately by extracting and counting its yttrium-90 daughter, while strontium-89 is estimated by difference from the total radiostrontium measurement. Appropriate corrections are made for self-absorption and counter efficiency at all stages. Calcium is determined by flame photometry.

⁴ Data from Radiation Protection Programs, Vol. 2, No. 12:25-30 Radiation Protection Division, Canadian Department of National Health and Welfare (Dec. 1964).

Cesium-137 is determined by gamma spectroscopy using a scintillation crystal and a multichannel pulse height analyzer. A sample consisting of 4.5 liters of milk is placed in a sample tray constructed in the form of an inverted well to accommodate the 5 x 4-inch sodium iodide crystal detector. The sample is counted for 100 minutes and the gamma spectrum recorded. Estimates are made of the potassium-40 and cesium-137 content of the milk by comparison of the spectrum with the spectra of standard preparations of these two radionuclides. With this method the potassium-40 concentration is determined and the Compton contribution of this radionuclide to the cesium-137 photopeak is subtracted to obtain the cesium-137 concentration. The stable potassium content is estimated from the potassium-40 concentration.

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In the iodine and strontium determinations, tests indicate that the statistical error (95 percent confidence level) in the chemical operations involved is about plus-or-minus 10 percent. This value is independent of the concentration of the radioisotope in the milk because it depends only on the recovery of the carrier. In the determination of cesium-137 this factor is not involved.

The chemical procedures error must be combined with the counting error which depends primarily on the concentration of the nuclide in the sample, the background radiation, and the length of time the sample and background are counted. This counting error has been evaluated mathematically for the particular counting arrangement used.

The overall errors, estimated on the basis indicated above, are given in table 9.

TABLE 9.—TOTAL ERROR FOR VARIOUS RADIONUCLIDE CONCENTRATIONS IN MILK *

Nuclide	Error for 10	Error for 50	Error for 100
	pc/liter	pc/liter	pc/liter
Strontium-89_	±25%	±20%	±15%
Strontium-90_	±15%	±10%	±10%
Iodine-131_	±50%	±20%	±10%
Cesium-137	±60%	±20%	±10%

a All errors are 2σ values, representing 95 percent confidence levels

Results

Table 10 presents monthly averages of strontium-90, cesium-137, and stable calcium and potassium in Canadian whole milk. Spot checks for iodine-131 and strontium-89 indicate that all samples had < 5 pc/liter.

TABLE 10.—RADIONUCLIDES IN CANADIAN WHOLE MILK, OCTOBER 1964

[Radionuclide concentrations in pc/liter]

Station	Calcium	Potassium	Strontium-	Cesium-
	(g/liter)	(g/liter)	90	137
CalgaryEdmontonFt. WilliamFredericton	1.20	1.6	27.9	93
	1.16	1.6	23.6	94
	1.16	1.7	47.2	172
	1.16	1.7	35.1	179
Halifax	1.14	1.7	52.3	218
	1.14	1.6	29.9	138
	1.16	1.6	25.4	108
	1.14	1.7	45.2	214
ReginaSt. John's, NildSaskatoonSault Ste. Marie	1.12	1.7	31.4	104
	1.12	1.6	40.4	177
	1.12	1.6	30.7	100
	1.18	1.6	37.6	162
Toronto Vancouver Windsor. Winnipeg	1.23	1.6 1.6 1.6 1.5	14.6 44.2 13.5 29.9	77 264 67 128
Average	1.16	1.6	33.1	143

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APPLICATION OF RADIONUCLIDE CONCENTRATIONS IN MILK TO DOSE ASSESSMENT, DECEMBER 1963-NOVEMBER 1964

Division of Radiological Health, Public Health Service

The concentrations of specific radionuclides in milk analyzed as part of the Pasteurized Milk Network (PMN) are reported on a monthly basis in *RHD*. In terms of radiological health surveillance activities, an important aspect of these data is the estimation of resultant radiation doses to population groups.

Approximate relationships between certain radionuclide intakes and dose have been applied to the formulation of daily intake guides (1) and permissible concentrations in selected environmental media (2). Although these guides are not themselves directly applicable to world-wide fallout, a comparison with environmental contamination levels does yield a measure of population dosage. In general, intakedose and dose-biological effect relationships used in formulating the guides cited are based on continuous intake over an entire lifetime. However, for general surveillance purposes, yearly average intakes, used with discretion, may be compared directly with the levels adopted as lifetime intake guides. Thus, the radionuclide concentrations in milk, averaged over a year's time, together with milk consumption data, might be used in conjunction with the references cited above to approximate the radiation dose to a specific population group from a specific radionuclide. Table 1 presents annual averages of radionuclide concentrations in milk sampled by the PMN. Limited data are available for estimating the average daily milk consumption (on a volume basis) for specific age groups in the U.S. population (3,4).

Total dietary intake is of prime interest, and since the intake via milk consumption constitutes only a portion of the total radionuclide intake, the relationship of milk intake to total dietary intake is of importance in evaluating milk surveillance data. The Federal Radiation Council (5) notes: "A number of studies have shown that conservative estimates of the strontium-90 to calcium ratio in the total diet may be made by multiplying the ratio of strontium-90 to calcium in milk in a particular locality by 1.5". Thus, a rough index of the total dietary intake of strontium-90 on an annual basis may be made from PMN annual averages by using this factor and the assumptions of approximately 1.2 g of calcium per liter in PMN samples and a 1.0 g daily intake of calcium.

In the case of iodine-131, milk can be considered the major source of intake because of the rapid distribution and consumption of fresh milk. With most other foods, normal processing and distribution allow time for the radioactive decay of this short-lived nuclide to insignificant levels.

The situation with respect to strontium-89 is more complicated. Its half-life of about 50 days makes it difficult to estimate the relative contribution made by sources other than milk to the total dietary intake. However, since the time from production to consumption of milk is generally much shorter than for most other food products, the short half-life of strontium-89 would cause milk to contribute relatively more of the total intake of strontium-89 than of strontium-90.

¹ This ratio may vary from 1 to 2, depending on changes in rate of fallout deposition and relative consumption of non-milk products whose contamination reflect temporal and local deposition patterns (6).

TABLE 1.—AVERAGE RADIONUCLIDE CONCENTRATIONS IN MILK, FOR THE TWELVE MONTH PERIODS NOVEMBER 1963-OCTOBER 1964* AND DECEMBER 1963-NOVEMBER 1964*

[pc/liter]

	All the safety and the safety of	Stronti	um-89	Stront	ium-90	Iodin	e-131	Cesius	m-137
	Sampling locations	Nov 1963- Oct 1964	Dec 1963- Nov 1964	Nov 1963- Oct 1964	Dec 1963- Nov 1964	Nov 1963- Oct 1964	Dec 1963- Nov 1964	Nov 1963- Oct 1964	Dec 1963- Nov 1964
Ala:	Montgomery	3	3	23 23	23	0	0	84	80
Alaska: Ariz:	Palmer	6 3	3 6 3		21	0	0	141	130
Ark:	PhoenixLittle Rock	4	3	45	44	0	0	25 129	123
Calif:	Sacramento	3 4	3 3	8	8	0	0	48	4
C. Z:	Cristobal	3	3	5	5	0	0	50	51
Colo:	Denver	4	4	10	19	0	0	91	8
onn: Oel:	Hartford	3	3 3 3	22 23 19	21	0	0	148	14
). C:	Wilmington	3	3	19	23 19	0	0	123 86	11
fla:	Tampa	3	3	16	15	ő	ő	236	23
la: Iawaii:	Atlanta	4 3	3 3	31	31	0	0	136	13
daho:	HonoluluIdaho Falls	11	10	11 27	12 27	0	0	78 168	16
11:	Chicago	3	10 3 3	19	19	0	Ö	115	11
nd:	ChicagoIndianapolis	3	3	21	21	0	1	98	9
owa: Kans:	Des Moines	6 5	5	25 21	25 21 30	0	0	91 66	8
Ky:	Louisville	4	3	31	30	0	0	91	8
faine:	New Orleans Portland	4 3	4 3	25 21 31 51 31	51 30	0	0	144 201	13 19
Md:	Baltimore	3	3			0	0	99	96
Mass:	Boston	3	3	22 33	22 32 18	0	ő	223	21
Mich:	Detroit	3	3	19	18	1	2	113	10
dinn:	Minneapolis	3 9	3 7	21 33	21 32	0	1 0	124 147	12 14
Miss:	Jackson Kansas City	5	4	41	41	0	0	107	10
Mo:	Kansas City	7	6	28	27	0	0	81	7
Mont:	St. Louis Helens	5	5 5	23 26	23	0	0	82 175	7
Vebr:	Omaha			26	27 23 25 26	0	ó	95	16
Nev:	Las Vegas	5	5 5	10	10	1	1	73	9
V. H:	Manchester	3 3	3 3 3 3 43	32 19	30	0	1	236	22
N. J: N. Mex:	TrentonAlbuquerque	3	3	11	19	0	1	114 54	110
V. Y:	Buffalo	3	3	20 26	11 20	0	Ô	141	13
	New York	• 3	3	* 20	25	0	40	159	15
	Syracuse				4 20	• 0		* 139	10
N. C: N. Dak:	Charlotte	14	3 12	36 57	36 56	0	0	112 152	10
Ohio:	Cincinnati	3	3	23	23	Ö	0	92	9
Okla:	Cleveland Oklahoma City	3	3 3 3	21 23	23 21 23	0	0	113 70	111
Ore:	Portland			31	30	0	0	164	15
Pa:	Philadelphia	3 3	3 3	20 29	20 29	0	0	119	11
n n.	Pittsburgh	3	3	29	29	0	0	148	14
P. R: R. I:	San Juan Providence	3 3	3 3	12 25	12 24	0	0	73 163	15
S. C:	Charleston	3	3	32	32	0	0	127	12
. Dak:	Rapid City	8	6	43	41	0	0	152	14
l'enn:	Chattanooga	5	4	41	32	0	0	120	11
Tex:	Austin		3 3 3	32 9 20	9	0	0	120 74 41	11 7 3 6
	Dallas						0	86	
Utah: Vt:	Salt Lake City	. 8	6 3 3 7	27 26	25	0	0 2	178 178	17
Va: Wash:	Norfolk	3	3	26 18	18	0	0	94	9
Wash:	SeattleSpokane	8 3 3 7 8	7 6	26 29	25 26 18 26 28	0	0	152 142	17 17 9 14 13
W. Va:	Charleston			27		0	0	81	7
Wis:	Milwaukee	3 8	3 3 8	18	26 18	0	1	130	12
Wyo:	Laramie	- 8		21	20	0	0	107	10
Vetwork	average	4	4	24.5	24.1	<10	<10	118	11

Note: Annual averages for Barium-140 at each station were <10.

* Annual averages were computed on basis of 53 weekly averages.

* Annual averages were computed on basis of 52 weekly averages.

* Annual averages were computed on basis of 49 weekly averages.

* Annual averages were computed on basis of 48 weekly averages.

No samples were collected during July 1964.

No samples were collected during July 1964.

The relative contribution of milk to the total dietary intake of cesium-137 is not well defined and depends principally on the amount of freshly deposited cesium-137 on products used for human and animal consumption, and the progress of cesium-137 through the food chain.

The data in table 1 are calculated as follows: results from all samples collected in each week (Sunday through Saturday) are averaged, and the averages for all weeks terminating in each of twelve consecutive months are averaged to obtain the annual average.2 To obtain the annual average daily intake (pc/day) of radionuclides from milk, the annual average concentration values (pc/liter) in table 1 must be multiplied by the annual average daily consumption (liters/day) of milk.

Monthly variations of radionuclide concentrations in milk are influenced by a number of combined causes such as meteorologic conditions and dairying practices, apart from considerations of original sources of radionuclides. The moving yearly average (table 1), obtained by updating the previous twelvemonth average by one month, shows variations averaged over the year and tends to minimize purely seasonal variations. This method, therefore, shows trends over a considerable period of time.

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sumption of selected food items in United States households, July 1962. Rad Health Data 4: 124-27. 5. Federal Radiation Council. Estimates and evaluation of fallout in the United States from nuclear weapons

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² Beginning with the October 1963 data, iodine-131 values of <10 pc/liter are considered to be zero for averaging purposes; previously, 5 pc/liter was used for calculating the averages.

STRONTIUM-90 IN TRI-CITY DIETS, MAY-JULY 19641

Health and Safety Laboratory, AEC

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Since March 1960, the Health and Safety Laboratory, through its quarterly diet study, has made estimates of the strontium-90 content of the average diet of individuals living in New York City, San Francisco, and Chicago.

Selected foods representing 19 food categories are purchased at each of these three cities about every 3 months and are analyzed for strontium-90. Using data from the U. S. Department of Agriculture (1) the annual consumption by an average individual can be grouped into the same 19 food categories. The annual dietary intake of strontium-90 can be estimated by summing the contributions from each category. Some food types are assumed to be representative of larger food categories, such as liquid milk for dairy products in general.

The consumption data (1) are based on a weight-as-purchased basis. Before the food samples for the Tri-City Diet Study are ashed for radiochemical analysis, they are prepared to a certain degree as if for actual consumption. For example, fruits are peeled, eggs are shelled, and poultry is boned. Therefore, concentrations of radioactivity in foods reported in the Tri-City Diet Study are based on the trimmed weight. No correction is made for the waste.

After two samplings at each city it was found that the calcium content of most food categories did not vary among cities, nor did it vary significantly with time. Calcium analyses of dietary components were performed for the third time recently, and further confirmed this result (2). Calcium analyses were therefore

discontinued and average calcium content of foods was computed and used to estimate the average annual intake of this mineral. Details of the sampling system and a discussion of the results obtained have been summarized (3).

Results obtained from the May-July 1964 sampling are presented in table 1. The variation with time of the daily intake of strontium-90 in the three cities is plotted in figure 1.

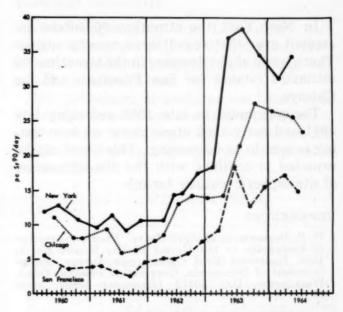


FIGURE 1.—DAILY INTAKE OF STRONTIUM-90
IN TRI-CITY TOTAL DIETS

Discussion

The previously noted geographic distribution pattern of strontium-90 in the diet is seen to persist as it has since the initial samplings in 1960: Levels have been highest in New York City and lowest in San Francisco. Due in part to its high annual consumption, milk continues to be the predominant source of strontium-90 in diet.

¹ Data from Fallout Program Quarterly Summary Report, HASL-155:208-10, Office of Technical Services, Department of Commerce, Washington, D.C. 20230 (January 1, 1965) price \$4.00.

Table 1.—AVERAGE PER PERSON DIETARY CONSUMPTION AND STRONTIUM-90 INTAKE,
-MAY-JULY 1964 SAMPLING

Food category	Averag		New You May		Chies July 1		San Fran June 19	
purchase and extend the manager	diet (kg/yr)	Calcium (g/yr)	pc/kg *	ре/уг	pc/kg *	pe/yr	pc/kg *	ре/уг
Bakery products Whole grain products Eggs Fresh vegetables Root vegetables Milk Poultry Fresh fish Flour Macaroni Rice Meat Shellfish Dried beans Fresh fruit Potatoes Canned fruit Fruit juices	37 11 16 43 17 221 17 8 43 3 73 73 1 3 68 45 26	37.0 10.0 9.1 15.0 6.1 234.3 9.2 10.8 8.6 0.7 1.1 10.9 0.8 2.9 12.6 5.8 1.3	29.8 ±1.6 87.9 ±2.3 6.5 ±0.2 20.3 ±0.5 8.6 ±0.5 32.0 ±0.8 2.8 ±0.2 0.6 ±0.1 32.6 ±0.6 16.6 ±0.5 4.4 ±0.2 1.7 ±0.1 36.0 ±2.2 8.0 ±0.3 7.8 ±0.7 2.8 ±0.7 2.8 ±0.3 3.2 ±0.3	1103 967 104 873 146 7,072 48 5 1,402 13 124 3 108 544 351 73 61	40.3 ±1.5 83.3 ±1.9 9.8 ±0.2 13.3 ±0.5 7.0 ±0.3 16.2 ±0.6 3.5 ±0.8 2.1 ±0.1 21.2 ±0.4 28.0 ±0.5 6.1 ±0.2 2.2 ±0.1 1.4 ±0.1 35.6 ±2.0 4.2 ±0.2 3.1 ±0.3 3.6 ±0.1 5.6 ±0.2	1491 916 157 572 119 3,580 60 17 912 84 18 161 107 286 140 94	26.3±1.5 50.9±2.1 1.8±0.1 22.4±0.6 6.9±0.6 8.7±0.5 2.2±0.1 0.6±0.1 12.7±0.4 12.4±0.5 0.4±0.1 2.1±0.1 4.5±0.2 13.6±1.6 0.4±0.1 2.6±0.4 1.5±0.4 1.5±0.4 1.5±0.4 1.5±0.4 1.5±0.4 1.5±0.4	973 560 28 963 117 1,923 37 5 546 37 1 153 4 41 27 162 39 80 72
Canned vegetables	20	4.2	6.7±0.6	134	10.9±0.5	218	3.6±0.5	72
Annual intake	674	383	19.6	13,181	13.4	9,039	8.6	5,768
pc Sr ⁹⁰ /g Ca in total diet				34.4		23.6		15.

D

In New York the strontium-90 intake increased over that found three months earlier. There was a slight decrease in the strontium-90 estimated intakes for San Francisco and for Chicago.

The maximum in late 1963 reflecting late 1961 and early 1962 atmospheric nuclear testing is seen to be decreasing. This trend can be expected to continue with the discontinuation of atmospheric nuclear testing.

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Recent coverage in Radiological Health Data:

Period	Issue
Thirteenth sampling (May-	
July 1963)	March 1964
Fourteenth sampling (August-	
October 1963)	June 1964
Fifteenth sampling (November	~
1963-January 1964)	September 1964
Sixteenth Sampling (February	D 1
1964-April 1964)	December 1964

^{*} Error terms are one standard deviation (due to counting).

ESTIMATED DAILY INTAKE OF RADIONUCLIDES IN CALIFORNIA DIETS, JANUARY-JUNE 1964

Bureau of Radiological Health California State Department of Public Health

Since January 1964, the Bureau of Radiological Health, California State Department of Public Health, has made estimates of radionuclide levels in the diets of Californians (1).

Recognizing that a "standard" or "typical" diet does not exist due to variations in individual tastes, an effort was made to select a diet which was reasonably representative of the food consumed in a given area. This objective was met by utilizing the "house" diet of a hospital in each of the 20 geographic areas of interest.

Hospitals were chosen as the source of diet samples under the hypothesis of their being as "reasonably representative" as any other source. General hospitals exist in each of the selected geographic areas and operate with trained dietitians. There is good reason to believe that hospitals utilize foods which are marketed in their respective communities. Also, working relations for entry into hospitals existed through the State Bureau of Nutrition and Hospitals.

Sampling procedure

In general, the sampling procedure is the same at each hospital. Samples are collected once every two months at each facility. Each sample represents the edible portion of a regular meal (the standard diet) for a full 7day week (21 consecutive meals).

After each sample is collected, it is suitably preserved and shipped to the Sanitation and Radiation Laboratory. Accompanying each sample is a record prepared by the dietitions indicating the types and quantities of food included.

Analytical procedures

After weighing at the Laboratory, each sample is homogenized, dried, and ashed prior to stable calcium, potassium, strontium, and sodium determinations.

The radiochemical analyses are based upon three basic procedures: chemical separations of strontium-89 and strontium-90; radium-226 analysis; and gamma spectroscopy of the entire wet sample.

TABLE 1.—ESTIMATED DAILY INTAKE OF RADIONUCLIDES IN CALIFORNIA DIETS,* JANUARY-FEBRUARY 1964

	Kg/capita	-		pe	c/capita/da	У				grams/capi	ta/day e. d	
City	per day b	Sr-89	Sr-90	Ra-226	Cs-137	Zr-95	Mn-54	Ce-141, Ce-144	к	Na	Ca	Sr
Bakersfield Berkeley Bishop Brawley Crescent City	2.1 2.5 1.0 2.7 2.9	8.1 1.5 6.2 4.8 24.0	12 17 19 7 70	0.7 0.5 2.6 1.0	54 77 63 39 173	48 80 35 44 67	1.2 8.2 5.1	0.8 5.8 2.0 7.5 11.0	2.7 3.5 2.3 3.6 3.7	3.6 4.0 2.5 4.7 4.0	0.8 0.6 0.7 1.2 1.5	0.00 0.00 0.00 0.00 0.00
Eureka Fresno Los Angeles Needles Quincy	3.1 2.7 1.9 1.3 2.5	3.9 5.6 0.7 0	17 5 15 15	3.1 2.3 2.5 1.3 0.8	79 41 73 88	17 41 3.7 51	4.6 13 — 1.6	7.0 6.5 2.5 2.4	2.3 3.7 1.2 2.7 1.7	3.0 5.2 3.5 4.8 4.1	1.2 0.7 1.0 1.1	0.00 0.00 0.00 0.00
Redding Sacramento Salinas San Bernardino San Diego	1.5 2.7 2.4 2.6 2.1	2.2 1.2 3.5 0.6 0.3	2.3 21 11 15	0.6 0.8 0.1 1.2 1.6	41 81 144 	31 49 40 —	3.1 9.3 4.6 3.4	1.0 0.8 0.4 3.6	2.2 3.2 3.1 1.9 3.1	5.3 4.7 2.6 4.9 7.8	0.9 1.2 0.7 0.9 1.1	0.00 0.00 0.00 0.00 0.01
San Luis Obispo Santa Barbara Santa Rosa Susanville Ukiah	1.7 3.0 2.0 2.1 2.2	0.4 3.0 2.0 7.9 4.9	2 6 13 11 15	1.5 0.2 0.3 1.0	28 59 80 55	31 69 41 54	5.2 - 2.2 13	4.2 3.8 3.1 2.4	2.0 1.6 2.8 1.4 3.1	3.1 2.2 2.7 3.6 2.6	1.2 0.4 1.0 0.8 1.7	0.00 0.00 0.00 0.00 0.00

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<sup>Based on analyses of hospital standard diets located in listed cities.
Kilograms of food per person per day in this diet.
Natural potassium contains 0.0119 percent of radioactive K-40.
Stable strontium consists of Sr-88-82.7 percent; Sr-87-7.0 percent; Sr-86-9.8 percent; Sr-84-0.5 percent; total-100.0 percent.
Dash indicates sample lost in laboratory procedures.</sup>

Data and discussion

The resultant estimates of daily intake of radionuclides in the California diets are given in tables 1-3 for January through June 1964 (2).

It should be noted that levels of radioactvity were observed to be far below those levels for which consideration should be given to protective health action.

TABLE 2.—ESTIMATED DAILY INTAKE OF RADIONUCLIDES IN CALIFORNIA DIETS, a MARCH-APRIL 1964

think, while still	Kg/capita per day b	No. Int.	10/4	po	c/capita/da	y			grams/capita/day e. d			
City	per day b	Sr-89	Sr-90	Ra-226	Cs-137	Zr-95	Mn-54	Ce-141, Ce-144	К	Na	Ca	Sr
Bakersfield Berkeley Bishop Brawley Crescent City	2.2 2.6 1.4 2.6 2.3	8.0 1.1 2.0 3.3 5.5	13 27 8 12 68	2.7 1.0 0.9 2.0 2.9	108 86 51 27	52 44 28 44	3.3 7.1 0	7.0 222	3.3 3.2 1.8 3.0 3.2	4.6 4.9 2.4 4.4 4.1	1.3 1.4 0.5 1.4 1.2	0.07 0.07 0.04 0.08 0.08
Eureka Fresno Los Angles Quincy	2.1 2.3 2.7 1.3	6.3 0 0 3.3	38 33 14 7	0.7 1.3 0.2 0.9	80 87	52 56	8.9	18_0	2.5 3.5 2.8 1.8	3.1 5.5 3.7 2.8	1.3 1.3 0.8 0.5	0.07 0.08 0.06 0.06
Redding Sacramento Salinas San Bernardino San Diego	2.2 2.6 2.3 1.9 2.2	10 0 4.2 8.0 3.5	14 15 18 11 2	2.4 1.4 0.9 0.6 0.3	55 85 55 48 45	35 36 39 36 42	-0 0 2.7 0.7	95 2.6 1.6 3.8 0	2.8 4.8 2.9 3.0 2.8	4.3 8.3 3.6 4.1 3.6	0.9 1.7 0.8 1.0 0.7	0.06 0.13 0.06 0.07 0.07
San Luis Obispo Santa Barabara Santa Rosa Susanville Ukiah	1.6 1.9 2.5 2.3 2.1	1.7 0 0 3.9 0.6	10 15 27 14 14	0.8 0.6 0.6 0.6 1.5	38 62 77 108 67	33 31 34 52 41	8.3 0 0 3.3	6.9 0 0 0	2.0 2.3 3.1 2.3 3.0	3.2 3.1 3.3 3.1 2.7	0.7 1.0 1.4 0.8 0.9	0.00 0.00 0.00 0.00

Based on analyses of hospital standard diets located in listed cities.
 Dash indicates sample lost in laboratory procedures.

TABLE 3.—ESTIMATED DAILY INTAKE OF RADIONUCLIDES IN CALIFORNIA DIETS,^a MAY-JUNE 1964

	Kg/capita			po	/capita/da	У			grams/capita/day e. d			
City	per day b	Sr-89	Sr-90	Ra-226	Cs-137	Zr-95	Mn-54	Ce-141, Ce-144	К	Na	Ca	Sr
Bakersfield Berkeley Bishop Brawley Crescent City	2.2 2.9 1.4 2.4 2.2	3.1 7.0 3.5 0	13 14 9.0 41 38	0.5 1.2 0.4 0.1 0.8	64 60 37 75 93	54 37 27 3.0	0 0 0.4 b	28 34 30 87 72	3.0 3.9 2.0 3.3	4.1 5.3 2.1 4.3	1.2 1.1 0.6 1.0	0.00 0.00 0.00
Eureka Fresno Los Angles Needles Quincy	2.3 2.3 2.7 2.3 1.8	0 0.2 0 0	28 11 15 17	1.1 0.6 0.8 0.6 0.6	92 65 61 63 103	28 37 34 47 27 144	1.6 0 10 106	0 0 18 0	2.6 2.8 3.0 3.0 3.0 2.3	3.3 3.2 4.2 3.2 3.6 4.6	0.9 1.2 0.9 0.8 1.0 0.4	0.0 0.0 0.0 0.0 0.0
Redding Sacramento Salinas San Bernardino San Diego	2.5 3.4 2.4 2.1 2.0	$1.0 \\ 13 \\ 0 \\ 0.4 \\ 0$	27 17 27 11 20	1.0 1.8 0.6 0.9 1.6	100 85 92 52 65	39 41 9.3 39 30	20 12 3.2 0.5 1.1	72 11 8.2 0	3.3 3.9 3.4 3.2 2.8	5.7 6.8 5.7 3.8 3.8	1.1 1.3 0.7 0.9 1.1	0.0 0.0 0.0 0.0
San Luis Obisco Santa Barbara Santa Rosa Susanville Ukiah	1.4 1.8 2.4 2.3 2.3	0 0 0 0	14 14 36 17 20	0.8 1.4 2.0 0.7 0.6	51 39 95 80	44 32 35 32	$\frac{0}{0}$ 4.3	72 0.5 68	1.9 2.5 2.8 3.0 3.1	3.8 3.6 3.3 3.2 3.9	0.6 0.8 0.9 0.8 1.5	0.0 0.0 0.0 0.0

Based on analyses of hospital standard diets located in listed cities.
 Dash indicates sample lost in laboratory procedures.

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Section III—Water

Gross Radioactivity and Strontium-90 in Surface Waters of the United States, September 1964

for tec-

0.07 0.04 0.08 0.06 0.07 0.08 0.06 0.02 0.06 0.13 0.06 0.07 0.07

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ata

Division of Water Supply and Pollution Control, Public Health Service

Levels of radioactivity in surface waters of the United States have been monitored by the Public Health Service Water Pollution Surveillance System since its initiation in 1957. Beginning with the establishment of 50 sampling points, this system has expanded to 131 stations as of February 1, 1965. These are operated jointly with other Federal, State, and local agencies, and industry. Samples are taken from surface waters of all major U.S. river basins for physical, chemical, biological, and radiological analyses. These data can be used for evaluating sources of radioactivity which may affect specific domestic, commercial, and recreational uses of surface water. Further, the system provides background information necessary for recognizing pollution and water quality trends and for determining levels of radioactivity to which the population may be exposed. Data assembled through the system and exact locations of sampling points are published in annual compilations (1-7).

Sampling procedures

The participating agencies collect one-liter "grab" samples each week and ship them "as is" to the Surveillance System Laboratory in Cincinnati for analysis. Gross alpha and gross beta radioactivity determinations on the suspended and dissolved solids are performed as frequently as deemed necessary.

Presently, gross alpha and beta determinations are made on either monthly composites of the weekly samples or on each weekly sample. Weekly alpha and beta determinations are scheduled for stations located downstream from known potential sources of radioactive waste. Weekly analyses are also conducted at all newly established stations for the first year of operation.

Normally, samples are counted within two weeks following collection or within one week after compositing. The decay of activity is followed on each sample for which the first analysis shows unusually high activity. Also, if a recount indicates that the original analysis was questionable, values based on recounting are recorded. All results are reported for the time of counting and are not extrapolated to the time of collection.

Analytical methods

The analytical method used for determining gross alpha and beta radioactivity is described in the eleventh edition of "Standard Methods for the Examination of Water and Wastewater" (8). Suspended and dissolved solids are separated by passing the sample through a membrane filter (type HA) with a pore size of 0.45 micron. Planchets are then prepared for counting the dissolved solids (in the filtrate) and the suspended solids (on the charred membrane filter) in an internal proportional counter. Reference sources of U₃O₈, which give a known count rate if the instrument is in proper calibration, are used for daily checking of the counter.

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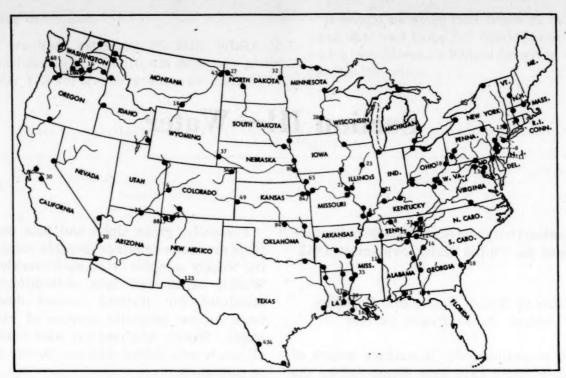


FIGURE 1.—SAMPLING LOCATIONS AND ASSOCIATED TOTAL BETA ACTIVITY (pc/liter) IN SURFACE WATERS,
OCTOBER 1964

Results

Table 1 presents September 1964 results of alpha and beta analysis of U.S. surface waters. The stations on a river are arranged in the table according to their relative location on the river, the first stations listed being closest to the headwaters. These data are preliminary. Replicate analyses of some samples as well as some analyses incomplete at the time of this report will be included in the system's "Annual Compilation of Data" (7). The figures for gross alpha and gross beta radioactivity represent either determinations on composite samples or means of weekly determinations where composites were not made. The monthly means are reported to the nearest pc/liter. When all samples have zero pc/liter, the mean is reported as zero; when the calculated mean is between zero and 0.5 the mean is reported as <1 pc/liter.

A geographical perspective of the radioactivity in surface water is obtained from the numbers printed near the stations as shown in figure 1 which gives the June 1964 average total beta activity in suspended-plus-dissolved solids in raw water collected at each station.

Strontium-90 determinations and results

Beginning in 1959, strontium-90 analyses of the total solids of surface waters were made quarterly on three-month composites of aliquots from weekly samples. Beginning in November 1962, the frequency of analysis was reduced to two quarterly samples per year at each sampling point except at those stations immediately below nuclear installations, where quarterly analyses were continued. The method used for determining strontium-90 is a modification of a procedure described by Harley (10). The yttrium-90 together with an yttrium carrier is precipitated as yttrium oxalate and the latter is washed and counted in a low-background anticoincidence, end-window proportional counter.

Table 2 presents the results of quarterly analyses of strontium-90 concentrations in U.S. surface waters for July-September 1964 as well as results for the previous three quarters for comparative purposes. The stations are arranged in the table according to their relative locations on the river, the first station being closest to the headwaters. Floyd and Weaver summarized the strontium-90 results

TABLE 1.—RADIOACTIVITY IN RAW SURFACE WATERS, SEPTEMBER 1964

[Average concentrations in pc/liter]

10 0 4E 100	Be	ta activi	ty	Al	pha activ	rity	The lotte A de A	Be	eta activi	ity	Alı	pha activ	ity
Station	Sus- pended solved Total		Total	Sus- pended solved Total		Total	Station	Sus- pended	Dis- solved	Total	Sus- pended	Dis- solved	Total
Animas River:				10.5	1.							7	
Cedar Hill, N. Mex.	4	14	18	1	1	2	E. St. Louis, Ill	8	19	27	1	2	3
Apalachicola River:						-	New Roads, La	13	22	35	1 2 1 0		3
Chattohoochee, Fla Arkansas River:	2	2	4	0	0	0	New Orleans, La		19	30	1	1	2
Coolidge, Kansas	6	43	49	0	12	12	Missouri River:	0	16	16	0	1	1
Ponca City, Okla	53	31	84	9	0	9	Williston, N. Dak St. Joseph, Mo	5	22	27	0	4	
Atchafalaya River:	00	31	01				North Platte River:	34	31	65	10	4	14
Morgan City, La	7	16	23	0	1	1	Henry, Nebr		0.	00			**
Bear River:							Ohio River:	3	34	37	0	21	21
Preston, Idaho	0	8	8	0	2	2	Cairo, Ill						-
Big Horn River:	-					7 11 44	Toronto, Ohio	0	6	6	0	1	1
Hardin, Mont	7	9	16	1	10	11	Platte River:	2	16	18	0	0	0
Chattahoochee River: Atlanta, Ga	5	9	14	-	0	<1	Plattsmouth, Nebr Potomac River:	58	22	80		3	
Columbus, Ga	2	7		<1	0	0	Washington, D.C.	08	22	80	12	3	15
Lanett, Ala	2	4	9	ŏ	ő	ő	Red River, North:	2	11	13	0	0	0
Chena River:							Grand Forks, N.	-	**	10		0	
Fairbanks, Alaska	1	8	9	0	1	1	Dak				1		
Clinch River:							Red River, South:	2	30	32	0	1	1
Clinton, Tenn	0	7	7	0	0	0	Alexandria, La						
Kingston, Tenn	12	23	35	. 0	0	0	Rio Grande River:	13	20	33	2	1	3
Colorado River:							El Paso, Tex						
Loma, Colo	4	0	4	1	17	18	Laredo, Tex		29	125	25	0	25
Page, Aris	1	34	35	0	5	5	San Joaquin River:	418	16	434	73	1	74
Parker Dam, Calif-	1	19	20	0	10	10	Vernalis, Calif San Juan River:	6	24	30		7	8
Ariz. Columbia River:		19	20	0	10	10	Shippoek N Mer	0	24	30	1		
Wenatchee, Wash	0	11	11	0	0	0	Shiprock, N. Mex Savannah River:	855	26	881	264	6	270
Pasco, Wash	123	981	1104	0	1	i	Port Wentworth, Ga.	000		001			
Clatskanie, Ore	18	150	168	<1	<1	<1	Snake River:	5	13	18	0	0	0
Connecticut River:							Wawawai, Wash						
Enfield Dam, Conn	0	9	9	0	0	0	South Platte River:	3	11	14	0	2	2
Cumberland River:							Julesburg, Colo				-		
Cheatham Lock,	0		8		0		Susquehanna River:	3	51	54	0	18	18
Tenn Delaware River:	0	8	8	0	0	0	Conowingo, Md Tennessee River:	2	9	11	0	0	
Philadelphia, Pa	1	7	8	0	0	0	Lenoir City, Tenn		9	11	0	0	
Escambia River:	-						Chattanooga, Tenn	1	7	8	0	0	
Century, Fla	5	8	13	1	0	1	Bridgeport, Ala	2		12		<1	< i
Great Lakes:							Pickwick Landing,	0	14	14	1	0	1
Duluth, Minn	0	1	1	0	0	0	Tenn				1		
Hudson River:							Tombigbee River:	2	11	13	0	<1	<1
Poughkeepsie, N.Y	4	13	17	0	0	0	Columbus, Miss		-		-		
Illinois River:	6	17	23		2		Wabash River:	3	8	11	0	0	1
Grafton, Ill Kansas River:	0	17	23	1	2	3	New Harmony, Ind Yellowstone River:	8	13	21	0	<1	<1
De Soto, Kans	51	33	84	7	1	8	Sidney, Mont		13	21	0	1	1
Maumee River:	0.	00	04	1 '			Juney, Montesses	25	18	43	3	3	
Toledo, Ohio	4	18	22	1	1	2					-	-	-
Mississippi River:			1				Maximum	855	981	1104	264	21	270
St. Paul, Minn	1	27	28	0	2	2							
			-				Minimum	0	0	1	0	0	(

Note: These data are preliminary; reanalysis of some samples may be made and additional analysis not completed at the time of the report may become available. For final data, one should consult the system's annual report.

obtained from 1959 through 1963 in the August 1964 issue of RHD (11).

Sixty-seven quarterly composite samples were analyzed for strontium-90 activity in July-September 1964. The arithmetic mean for these results was 2.8 pc/liter. The mean for 69 results of the previous quarter was 3.0 pc/liter. These values are essentially the same, although different sets of stations were sampled in each quarter.

The highest strontium-90 result for this quarter (8.2 pc/liter) occurred, as it did the previous quarter, at Grand Forks, North Dakota (Red River, North). While there are no standards for strontium-90 activity of total solids in surface water, the Public Health Service Drinking Water Standards set the limit for

strontium-90 concentrations in drinking water at 10 pc/liter (11). This limit for public water supplies is greater than the highest level observed during July-September 1964.

Discussion

The monthly dissolved beta activity averages exceeded 100 pc/liter at only two stations, the Pasco, Washington, station and the Clatskanie, Oregon, station on the Columbia River. The dissolved beta activity values observed were 981 and 150 pc/liter, respectively.

Dissolved alpha activity, associated with the dissolving of natural surface minerals by water, ranged from 0 to 21 pc/liter. Four stations had monthly average dissolved alpha activity greater than 10 pc/liter.

Table 2.—QUARTERLY AVERAGE STRONTIUM-90 CONCENTRATIONS IN SURFACE WATERS, OCTOBER 1963-SEPTEMBER 1964 [Concentrations in pc/liter]

Station	Oct Dec. 1963	Jan Mar. 1964	Apr June 1964	July- Sept. 1964	Station	Oct Dec. 1963	Jan Mar. 1964	Apr June 1964	July Sept 196
Allegheny River:					O-l- Nil-				
Pittsburgh, Pa	2.8		2.3		Omaha, Nebr	3.4	3.3	6.4	4.
Cedar Hill, N. Mex		1.7		1.5	St. Joseph, Mo. Kansas City, Kans	100	2.9	0.4	5.
Apalachicola River: Chattahoochee, Fla		1.5		1.6	Missouri City, Mo	4.1		9.4	
rkansas River:		1.5			St. Louis, Mo		3.1	1111	5.
Coolidge, Kans		0.9		2.5	Pittsburgh, Pa North Platte River:	3.3		2.1	
Fort Smith, Ark	4.4	3.5	6.0	4.8	North Platte River: Henry, Nebr	0.7		9.9	
Little Rock, Ark	4.4	0.0	4.2	1.0	Ohio River:	0.7		3.3	
Little Rock, Ark	4.4		4.4	11-	Toronto Ohio	2.8		2.3	
Atchafalaya River: Morgan City, La.*				4.0	Addison, Ohio. Huntington, W. Va. Cincinnati, Ohio. Louisville, Ky. Evansville, Ind.		2.2	0.0	3
lean Missens				4.0	Cincinnati. Ohio	3.0	1.6	2.2	3
Preston, Idaho		1.1		2.5	Louisville, Ky	4.0	1.0	2.6	
lig Horn River:					Evansville, Ind		2.5		3
Hardin, Mont	2.3		6.4		Cairo, Ill	2.8		3.4	
Sioux Falls, S. Dak	5.0	2.8		4.6	Bastrop, La.		3.0		2
Chattahoochee River:					Pend Oreille River: Albeni Falls Dam, Idaho				
Atlanta, Ga	1.7		2.4		Platte River:	1.3		1.2	
Lanett, Ala	•	1.7		1.8	Plattsmouth, Nebr		2.1		4
Chena River:			0.0		Potomac River:				
Fairbanks, Alaska	0.2		0.8		Williamsport, Md	1.7	1.4	1.3	,
Lewiston, Idaho		0.8		1.0	Great Falls, Md. Washington, D. C.		1.4		
Winch Disser.					Rainy River:				1
Kingston, Tenn	6.5	7.5	4.9	1.3	Baudette, Minn. International Falls, Minn.	3.8		5.0	
Clinton, Tenn		1.0		0.0	Raritan River:			0.0	
Lome Colo	1.4		2.5		Perth Amboy, New Jersey				
Page, Aris Boulder City, Nev Parker Dam, Calif-Aris	1.5	5.9	1.6	4.9	Perth Amboy New Jersey				
Parker Dam, Calif-Aris	1.5	1.9	1.0	2.2	Perth Amboy, New Jersey				
	1.1		1.0	112	Red River, North:		1		
Columbia River: Northport, Wash		1.5		2.1	Grand Forks, N. Dak	7.1	4.9	9.4	
Wenatchee, Wash	2.8	1.5	1.2	2.1	Red River, South: Denison, Tex		5.8		
Pasco, Wash McNary Dam, Ore Bonneville, Ore Clatakanie, Ore. Connecticut River:	3.4	3.1	1.2	2.5	Index, Ark Bossier City, La Alexandria, La Rio Grande River:	4.3		5.4	
McNary Dam, Ore	2.5	2.2	1.1	1.6	Bossier City, La.		4.4		1
Clatakanie, Ore	1.6	2.0	1.0	1.0	Rio Grande River:	4.1		4.4	
Connecticut River:	1.0		1		Alamosa, Colo El Paso, Tex Laredo, Tex Brownsville, Tex	0.8	1	1.9	
Wilder, Vt		1.3		1.4	El Paso, Tex		0.6		
Enfield Dam, Conn	1.8	1.7	1.5	1.9	Brownsville Tex	2.4	2.6	5.6	
Coosa River:					Rosnoke River:	1	0		
Rome, Ga.b.				1.7	John H. Kerr Resr/Dam, Va	2.6		1.5	
Cumberland River:		1			Sabine River: Ruliff, Tex		2.5	1	
Clarksville, Tenne Cheatham Lock, Tenne				1.7			2.0		
Cuyahoga River:	1				Courtland, Calif	1.0	1.0	1	
Čleveland, Ohio Delaware River:	1	4.3		4.5	Courtland, Calif	1	1.6		
Martina Creek. Pa	1.6		1.5				1.0		
Martins Creek, Pa		1.8		1.7	Vernalis, Calif	1.5		1.0	
Philadelphia, Pa	2.1		2.0		San Juan River:	2.1		2.9	
Escambia River: Century, Fla	1.2	1	1.8		Shiprock, N. Mex Savannah River:	1		2.0	
Lireat Lakes:					North Augusta, So. Car Port Wentworth, Ga		1.7		
Duluth, Minn	0.7	0.8	0.7	0.8	Port Wentworth, Ga	2.4	1.7	2.3	
Milwaukee, Wis-	0.8	0.8	1.2		Schuylkill River: Philadelphia, Pa.		1.4		
Gary, Ind		1.2		1.0					
Gary, Ind. Port Huron, Mich.	1.2		1.5		Berryville, Va	1.0		1.1	-
Detroit, Mich	5.0	1.4	2.7	1.6	Ship Creek: Anchorage, Alaska	1	0.4		
Green River:	1 0.0	1			Snake River:	1		1	
Dutch John, Utah		4.6		4.4	Ice Harbor Dam, Wash		1.0		
Hudson River: Poughkeepsie, N. Y	5.0		2.1		Wawawai, Wash Payette, Idaho South Platte River:	0.7		0.9	
Illinois River:					South Platte River:	- 0.0	1		1
Peoria III	2.3		3.8		South Platte River: Julesburg, Colo	1.8		1.6	
Grafton, Ill	-	3.8	'	3.4	Post Falls, Idaho		1.3		1
Winfield Dam, W. Va		1.1		1.8	Susquehanna River		1.0		
Kanasa River					Sayre, Pa		1.3		
De Soto, Kans	5.2		5.9	'	Conowingo, Md	3.0	1	1.3	1
Keno Ore	1.8		1.7		Lenoir City, Tenn	2.1		1.6	
Klamath River: Keno, Ore	1				Chattanooga, Tenn	2.1 2.2 2.2	2.0		
Cincinnati, Onio		3.3	1	2.4	Chattanooga, Tenn Bridgeport, Ala Pickwick Landing, Tenn	- 2.2	2.1	1.5	1
Maumee River:	2.7		4.2		Tombighee River				
Maumee River: Toledo, Ohio	1				Columbus Miss	-	3.1		
Lowell, Mass	-	1.8	3	2.6	Truckee River: Farad, Calif Verdigris River:				. 1
Mississippi River:	4 9	3.5	2	5.3	Farad, Calif	- 1.0	'	1.2	•
Mississippi River: St. Paul, Minn Dubuque, Iowa Burlington Lowa	4.3	0	5.2		Nowata, Okla	6.0		6.6	3
Burlington, Iowa E. St. Louis, Ill Cape Girardeau, Mo	3.8	2.1	3	4.4	Nowata, Okla	1			
E. St. Louis, Ill.	- 3.8	2.1	4.5	4.2	New Harmony, Ind	- 2.5	'	4.2	•
W. Memphis Ark	3.6		4.3	1	Willamette River: Portland, Ore	0.8	5	0.3	3
W. Memphis, Ark Vickaburg, Miss		2.	7	3.7	Yakima River:	1			
Delta, La	3.4		3.6		Yakima River: Richland, Wash Yellowstone River:	-	0.3		
Delta, La. New Roads, La. New Orleans, La.	-	2.1	3.6	3.7	Yellowstone River: Sindey, Mont		1.9		
Missouri River:						_			- -
Missouri River: Williston, N. Dak Bismarek, N. Dak Yankton, S. Dak	2.8		3.1	1 40	Maximum	- 7.1	7.5	9.4	
Bismarck, N. Dak	3.3	3.	4.8	4.0	Minimum	0.5	0.3	0.3	

The radioactivity associated with dissolved solids provides a rough indication of the levels which could occur in treated water, since nearly all suspended matter is removed by the treatment process (12). The Public Health Service Drinking Water Standards state that in the absence of strontium-90 and alpha emitters,1 a water supply is acceptable when the gross beta concentration does not exceed 1,000 pc/liter (11). Analysis of the preliminary September 1964 data indicates elevated levels of alpha and beta activity (pc/liter) in the suspended solids fraction. The monthly average suspended alpha and beta activities in the San Juan River at Shiprock, New Mexico, were 264 and 855 pc/liter, respectively. At El Paso, Texas, and Laredo, Texas, on the Rio Grande the suspended alpha activities were 25 and 73 pc/liter and the suspended beta activities were 96 and 418 pc/liter, respectively. The dissolved alpha and beta activities were not significantly different from those commonly observed at this sampling station. The samples associated with these unusually high averages contained large quantities of suspended solids. Thus, only quite small aliquots could be used in the laboratory determinations to avoid excessive beta self-absorption. In such cases the relatively large multiplication factor, together with the usual range of counting errors associated with standard counting periods, can result in apparently abnormal levels of radioactivity. The specific activity (pc/g) of the samples discussed above was determined and found

¹ Absence is taken here to mean a negligibly small fraction of the specific limits of 3 pc/liter and 10 pc/ liter for unidentified alpha emitters and strontium-90 respectively.

to be comparable to specific activities observed normally in samples from this station. The apparent increased activity, therefore, may be attributed to the extremely high suspended solids content of the sample and considered essentially of natural origin.

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² Single free copies of this publication may be obtained from: Public Inquiries Branch, Public Health Service, U.S. Department of Health, Education, and Welfare, Washington, D.C. 20201.

RADIOACTIVITY IN CALIFORNIA SURFACE WATERS, 1 JANUARY-JUNE 1964

Bureau of Radiological Health, State of California Department of Public Health

Gross beta results obtained by the Bureau of Radiological Health in its monitoring of California surface water during the period January to June 1964 are summarized below. The importance of this facet of the Bureau's environmental surveillance program stems from the fact that most of California's domestic water supplies come from surface sources. Radioactivity in such water supplies consists of the natural radioactivity in surface streams, any radioactivity that may be added by the discharge of sewage or industrial waste effluents. and radioactivity from fallout, particularly fallout into open terminal or distribution reservoirs. Most of the supplies sampled represent raw surface waters, (figure 1), although a few wells, along with some water supplies that use infiltration galleries, are also sampled.

It is necessary to monitor domestic water supplies on a continuing basis, since it is impossible to forecast levels of radioactivity in these supplies on the basis of radioactivity in rain, snow, or surface streams. The Bureau has established a monthly sampling schedule whereby 500-ml samples are collected and the total solids are analyzed for alpha and beta radioactivity. In addition, a three-liter sample is collected each month for a period of approximately six months to make up a composite of approximately five gallons for specific radionuclide analysis.

Analytical procedures

Radionuclide analyses of water are carried out in the Sanitation and Radiation Laboratory. All measurements of alpha and alphaplus-beta activities are made with windowless gas-flow proportional counters. Counting methods used are in accordance with U.S. Public Health Service recommended procedures (1).

Individual samples are evaporated down to dryness and the residue ashed at 450° C. The ashed sample is dissolved and transferred to an aluminum planchet for beta and alpha-plusbeta counting. Gamma emitting radionuclides are determined for composite samples. Indi-



FIGURE 1.—CALIFORNIA SURFACE WATER SAMPLING STATIONS

Data from 'Radiological Health News," Vol. 3 No. 3, July, 1964, State of California Department of Public Health, Bureau of Radiological Health, 2151 Berkeley Way, Berkeley 4, California.

TABLE 1.—GROSS BETA ACTIVITY IN CALIFORNIA SURFACE WATER. JANUARY-JUNE 1964

[Concentrations in pc/liter]

Sampling station	January	February	March	April	May	June
ntioch	*4	• 2	• 23	97	36	7-
learlake Highlands	*0	b	* 8	a 25		8
rescent City	• 3	• 0	.0		*8	4
Secondido	• 6	* 15	25	* 24	30	2
ort Bragg	- 5	*6	* 15	47 40	• 14	
Lake Millerton	• 15	25	26	30	32	7
dariposa	• 13	• 19	* 12	108	88	3
Metropolitan Water District of Southern				8 .	-	
Lake Havasu	30		* 17	=		1
Lake Matthews	36	* 10	* 19	* 12	30	• 1
Monterey	* 16	* 13	* 0	50	38	* 1
North Marin Water District	• 24	.9	* 26	56	69	- 4
Wyandotte Irrigation District	_	100 1 -	24	163	=	8.5
California Water Service		7			27	
lacerville	30	• 0	* 13		39	
ledding.	a 15		• 6	55	38	
acramento	* 18	* 10	* 5	30	34	
an Francisco:						
Alameda East	26	22	27	94	33	
Brightside Weir Calaveras Reservoir	• 6	-0	- 23	* 18	50	1
Crystal Springs Line No. 1	a. e 15	a. e 14	a. 0 12	* 42	* 43	
Crystal Springs Line No. 2	• 17	_	_	_	77	
Crystal Springs Line No. 2 Hetch Hetchy Lombard Reservoir	* 27	. 9	• 21	-	187	
Moccasin	_	_	a. a 15	* 21	_	
San Andreas Line No. 2	_	_	-	° 58	· 43	
an Jose	*0	.0	.0		39	
an Luis Obispo	* 16	• 13	*0		-	
Santa Barbara		-		42	* 15	
anta Cruz		30	* 10	* 17	• 21	
Santa Rosa	* 8	* 11	* 16	72	• 17	
Scotia	• 3	*14	*1	51	• 15	
Tahoe City	* 11	• 9	-	55	32	1 1 1 1
Vallejo:						
Fleming Hill Swanzy Reservoir	53 • 21	36	32	61 59	49 23	1.01
Vista			-	-	=	
Willits	35	30	* 10	104 72	34 54	1
Yosemite						
Maximum	53	36	32	163	187	1
Minimum	0	0	0	12	8	

<sup>The counting rate of the sample is not equal to at least twice the 95 percent statistical counting error but the value reported is the best available estimate.
Dash indicates no sample collected or analyzed.
Average of more than one sample for the month.</sup>

vidual samples are composited in small glass petri dishes. The petri dishes are counted for 100 minutes using a gamma scintillation spectrometer. The quantities of the radionuclides of concern are calculated through the computer solution of eight simultaneous linear equations. Analysis is complicated by the presence of 0.51 Mev gamma radiation of unknown origin which may arise from a radionuclide emitting approximately this gamma energy or annihilation of a positron, where the positron resulted from either direct emission or "pair production".

Discussion

Table 1 shows the monthly average beta activity in the suspended-plus-dissolved solids in raw surface water in California from January through June 1964. Following treatment, these waters are used for industrial and domestic purposes. Because alpha activity in water has in general been undetectable or very slight, alpha activity analyses are not presented. Little increase in the radioactivity level of surface water has been observed in spite of some increase in fallout.

Table 2.—RADIONUCLIDE CONCENTRATIONS IN CALIFORNIA SURFACE WATER, 1960-1963

.[Concentrations in pc/liter]

Sampling station and date	Ba-La ¹⁴⁰	K40	Znes	Zr-Nb95	C8137	0.51 Mev	Inst	Ce141_Ce144	Sr89	Sr90
Berkeley Jan. 61—June 61 July 61—Dec. 61 Jan. 62—June 62 July 62—Dec. 62	*58		*1	* 16 * 6	b* *	*1 * * *1		*1,220	22 *1 3	*0.1 *0.1
Chula Vista Feb. 62—July 62——— Aug. 62—Jan. 63————		*1 *1	*1	*8		2		183	6	0.7
Cresent City July 60—Dec. 60. Jan. 61—June 61. July 61—Dec. 61. Jan. 62—June 62. July 62—Dec. 62. Jan. 63—June 63.		* 18 * *2	:	* * 8 * 33 3	*	* * 15 *	:	*	1 1 2 *1	0.1 0 *0 *
Dos Palos July 60—Jan. 61 July 60—Oct. 61 Feb. 62—July 62 July 62—Jan. 63 Feb. 63—July 63		*		* 32 *3 23		* 6 *1 *1		* 2	2 8 *2	0 0 0 0
El Centro Oct. 60—March 61 April 61—Sept. 61 Oct. 61—March 62		* 4		99		* *1 2			44 48	1.5 0.6 *0.5
Eureka June 60—Nov. 60 Jan. 61—June 61 Aug. 61—Jan. 62 Jan. 62—July 62 Oct. 62—Dec. 62			٠	* 63 130 98	٠	* * 11 3	•	* *763 145	9 187	0.3 0.3 3.6 0.1
Fresno, Lake Millerton July 60—Jan. 61 Feb. 61—June 61 Aug. 61—Jan. 62 Feb. 62—July 62 Aug. 62—Jan. 63 Feb. 63—July 63	*	* *5	٠	* 44 22 3 51		* 20 1 * 1 4	٠	* * 296 *	3 4 2 *1	0.3 0.3 0.3 0.4
Los Angeles Dept. of Water and Power Jan. 61—July 61 July 61—Dec. 61. Jan. 62—June 62		*13		92 6		* *2		*8	8 5	*0.: 0.: 0.:
Metropolitan Water District of Southern Calif. Jan. 61—June 61 July 61—Dec. 61 Jan. 62—June 62 July 62—Dec. 62 Jan. 63—June 63		* 3 *1	* *1	297 2 152 6	:	* 13 * 6 1	:	* 1,970 *2 19	* 4 10 5	0 *0. 0.: 1.:
Monterey July 61—Dec. 61 Jan. 62—June 62 July 62—Jan. 63 Jan. 63—June 63		25 * 9	*1 • 2	119 15 27		18 *1		* 1 3	7 2 *1	0 0 •0.
Napa Feb. 61—July 61	:	* * 6	*1 • 9	2 56 *4 53	*	2 1 *1 11		*11 *	22 *4 3 *2	0.: 0.: *
North Marin Jan. 61—July 61 Aug. 61—Dec. 61 Jan. 62—July 62 July 62—Dec. 62 Jan. 63—July 63		11 34 *12 *7	. *1	16 23	:	* 14 2 4 2		:	7 5 9 14	*0.: 1.: 0.: 0.:
Redding Jan. 61—July 61 July 61—Dec. 61 Jan. 62—June 62 July 62—Dec. 62 Jan. 63—June 63		22 *3		9 16 191 140		* * *2 32 6		3 593	4 4 7 12	0. 0. 0. 0.
Sacramento Aug. 60—Feb. 61— March 61—Sept. 61 Sept. 61—Feb. 62 March 62—Aug. 62 Sept. 62—Feb. 63 Feb. 63—June 63		* 21 * 7	• 2	113 10 41 2		* 8 *1 3 3 3		*1 2	39 2 3 *1	0.: 0.: 0.: 0.: 0.:
San Diego Nov. 60—April 61 Aug. 61—Dec. 61 May 62—Oct. 62		7 *10 *6	*1	*10 46		* *1 *2		91	40 375	0.: 0.: 0.:

Table 2.—RADIONUCLIDE CONCENTRATIONS IN CALIFORNIA SURFACE WATER, 1960-1963—Continued

[Concentrations in pc/liter]

Sampling station and date	Ba-La140	K40	Zn ⁶³	Zr-Nb95	Cs137	0.51 Mev	Isst	Ce141_Ce144	Sr89	Sree
San Francisco Dec. 61—May 62	L-E00	*9	1.24	2	11/200	2	A //	-YETTEVY	8	0.4
San Luis Obispo May 61—Oct. 61 Jan 62—June 62	13	4		13 42		13 4		* 351	5	1.2
Santa Barbara June 61—Nov. 61 Dec. 61—June 62 July 62—Nov. 62		4		1 71 34		* 4 5		* 532 *2	6 5	*0.7 0.3 0.6
Santa Cruz Nov. 61—April 62			*2	643		6			154	0.6
Santa Rosa July 61—Dec. 61 Jan. 62—June 62 July 62—Dec. 62 Jan. 63—July 63	a salt di	*6	*1	10 1 457		6 *1 17		e Legin	3	0.2 0.3 1.1 0.3
Vallejo Jan. 62—July 62 July 62—Dec. 62 Jan. 63—July 63	and i	* 47	*2 • 13	51 27	* 3	2 5 9	7	3 5	7 13 3	0.6 0.6 1.0
Yosemite May 61—Oct. 61 July 61—Oct. 61 Nov. 61—April 62 Oct. 62—March 63 Jan. 63—July 63	:	:	* • 2	28 8 87 16	. 1	5 * 6		3,160	4 *3 *14	2.0 0.4 0.2 0.9

Asterisks are used to indicate that the counting rate of the sample is not equal to at least twice the 0.95 error. The value reported is the best available estimate but is not statistically significant. An asterisk alone indicates no activity was detected.

Mn⁵⁶ instead of Zn⁶⁵.

Table 2 shows specific radionuclide concentrations in California surface waters for 1960-1963 bi-annually by stations.

Table 3 shows maximum radionuclide concentrations in California surface waters for the period 1960-1963. Related values of the International Committee on Radiological Protection Standards (2) for continuous population ex-

TABLE 3.—MAXIMUM OBSERVED RADIONUCLIDE CONCENTRATIONS IN CALIFORNIA SURFACE WATERS AND RELATIVE ICRP STANDARDS

Nuclide	Observed maximum (pc/liter)	MPC (pc/liter)
Ba-La ¹⁴⁰	58	10,000
Zn ⁶⁵	34 2 643	10,000 20,000
Cs137 I 131	3	2,000 670
Ce ¹⁴¹ -Ce ¹⁴⁴	3,160 375	3,300 (Ce ¹⁴¹)
Sr	3.6	33

posure are also shown. No standards are given for potassium-40 which occurs naturally. It is observed that none of the six-month composite raw surface water samples had specific radionuclide concentrations exceeding their respective MPC. For each radionuclide reported the minimum concentration observed was below detectable levels.

Previous coverage in Radiological Health Data:

Period	Issue
1961-June 1962	April 1963
July-December 1962	September 1963
January-June 1963	March 1964
July-December 1963	September 1964

REFERENCES

- Division of Radiological Health. Radionuclide Analyses of Environmental Samples R59-6, Radiological Health Research Activities, Robert A. Taft Sanitary Engineering Center, Cincinnati 26, Ohio (November 16, 1959).
- 2. ICRP Report Number 2.

RADIOACTIVITY IN KENTUCKY WATERS, MAY 1963-JUNE 1964

Radiological Health Program Kentucky Department of Health

Radioactivity is usually present in all natural surface waters. Prior to 1942, any radioactive minerals in natural waters were there as a result of cosmic radiation and naturally occurring radionuclides in the earth's soil which had been taken up by these surface waters. After that time fallout from nuclear weapons testing and possibly radionuclides in waste water from the atomic industry operations could contribute some additional radioactivity. Sampling of surface waters in the Commonwealth of Kentucky is done for the purpose of establishing the levels of radioactivity presently existing and to establish any relationship between radioactivity levels and stream characteristics.

Surface water samples are collected from 48 sampling locations (figure 1) in seven river basins. General description of the basins in Kentucky can be found in "Surface Water Records of Kentucky" (1).

The suspended and dissolved solids in these samples are analyzed for gross alpha and beta radioactivity. Generally, no attempt has been made to determine the specific radionuclides present. Table 1 presents the results of the alpha and beta analyses of the water samples collected from May 1963 through June 1964.

The analytical method used for determining gross alpha and beta radioactivity is similar to that used by the PHS R. A. Taft Sanitary Engineering Center (2). One gallon samples are collected and shipped in polyethylene bottles. The volume of water analyzed is selected so that no more than 500 mg of either suspended or dissolved solids is present during counting. Generally, one liter of water is used. The sample is first filtered through a millipore filter Wattman 2/0 to separate the suspended and dissolved solids. The filter is then ashed. The filtrate is evaporated leaving the dissolved solids which are also subsequently dried.

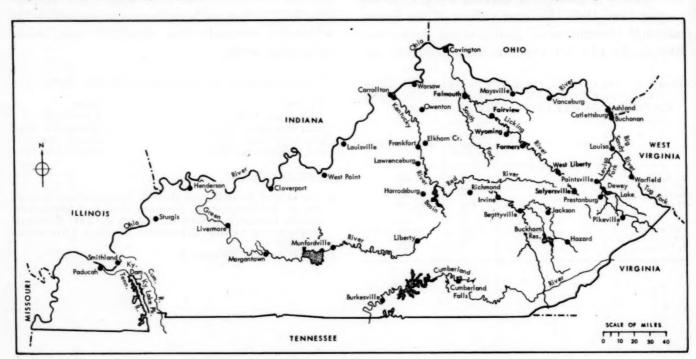


FIGURE 1.—KENTUCKY SURFACE WATER SAMPLING STATIONS

TABLE 1.—AVERAGE GROSS RADIOACTIVITY IN KENTUCKY WATER, MAY 1963-JUNE 1964

[Concentrations in pc/liter]

River basin and location	Dates	alan E	Beta		STATE OF THE	Alpha	
come (moreased furballence.	sampled	Suspended	Dissolved	Total	Suspended	Dissolved	Total
tis Sandy River			STYSEPHIN	aj neritri	000717-92	(CH - 50 150	O POR
Sig Sandy River Catlettsburg	6/5/64	6	9	15	<1		-
Buchanan	6/4/64	6 29	11	15 50	<1 <1	<1 <1	>
Louisa: Louisa Fork	6/4/64	12	10	22	<1	<1	<
Tug Fork	6/4/64	32	9	41	<1	<1	
Warfield	6/5/64 6/4/64	16	7	ii	<i td="" <=""><td>≥i </td><td>VVVV</td></i>	≥i	VVVV
Paintsville	6/4/64		10	26	<1 <1	<1	<
Prestonsburg	6/4/64	3	15	18	<1 <1	<1 <1	<
Dewey Lake	6/4/64 6/4/64	15	8 7	23 10	<1 <1	<1 <1	5
Cumberland River	in/teel				11		
Burkesville	6/30/64	9	17	26	<1	<1	
Smithland.	6/30/64	8 12	8 8	16	<1 <1	₹ <u>i</u>	2
Cumberland Falls	6/30/64	12	8	20	<1	<1	2
Green River Liberty	5/13/64	9	9	11	-1	-	11 201 1
Munfordville	5/13/64	13	9	11	<1 <1	<1 <1	111111111111111111111111111111111111111
Morgantown	5/13/64	13	10	- 22 17	<i< td=""><td><1 <1</td><td>5</td></i<>	<1 <1	5
Livermore	5/13/64	10	12	22	<î	<i td="" <=""><td></td></i>	
Henderson	5/13/64	4	15	19	<1	<1	***
Kentucky River Basin Hazard	0/0/04		10				
Jackson	6/8/64 6/8/64	15 7	19 20	27 35 28	0	0	
Buckhorn Reservoir	6/8/64	13	21	90	0 <1	<1 <1	and the f
South Fork	6/8/64	21	25	46	<i>1</i>	<i>1</i>	
Beattyville	6/8/64	16 15	18	33 27 34 39 57	<i< td=""><td><i td="" <=""><td></td></i></td></i<>	<i td="" <=""><td></td></i>	
Irvine	6/8/64	15	12	27	<1	<1	
Red River	6/8/64	17	17	34	<1	<1	
Richmond	6/8/64	26	13	39	<1	<1	
Harrodsburg Lawrenceburg	6/8/64	34	23	57	<1	0	
Frankfort	6/8/64	17 26 34 20 26	22	42 54	<1 <1	<1	
Frankfort Elkhorn Creek	6/8/64	8 43 36	13 23 22 28 35	43	0	0	
Owenton	6/8/64	43	25	68	<1	<1	
Carrolton		36	23	68 59	<i< td=""><td><i< td=""><td></td></i<></td></i<>	<i< td=""><td></td></i<>	
Licking River	- / /	101			1	for malley	
Salyersville	8/18/63	6	12	18	<1	<1	- 1
West Liberty Farmers		8 3	15 26	23	<1	<1	
Wyoming			19	18 23 29 24	<1 <1	<1 <1	
Fairview	8/18/63	14	31	45	<i< td=""><td><1</td><td></td></i<>	<1	
Fairview Falmouth	8/18/63	10		45 50	≥i	<i< td=""><td></td></i<>	
Ohio River			-				11111111
Ashland	- 7/31/63	3	14	17		1	
Vanceburg.	7/31/63	3 3	15	18	2	1	
Maysville	7/31/03	3	14	17	1	0	
Warsaw	7/31/63	3	14 12	17 15	1 0		
Carrollton	7/31/63	3	14	17	2		
Louisville	7/31/63	4	16	90	1	1	
West Point	7/31/63	4	19	23	1	Ō	
Cloverport	- 7/31/63	. 8	20	28	1	0	
HendersonSturgis		15		23 28 33 26	1	1	
Tennessee River					1	1	
Paducah	7/25/63	3 7	18	25	0	0	
Kentucky Lake Park	7/25/63	4	16	20	<1		
Kentucky Dam	7/25/63	15	20	35			

The solid samples are counted for alpha and beta activity in a proportional counter.

Although the few samples collected at irregularly spaced time intervals provide only an indication of base-levels of radioactivity, general levels of gross beta radioactivity have been established for the 48 stations. In addition, some determination of gross alpha concentrations, half-life values, gamma energy spectra, and specific isotope concentrations have been made to establish the probable source of the higher concentrations of gross

radioactivity. In specific instances, data on stream flow at the time of collection, and the quantity of suspended and dissolved solids was obtained.

From this initial sampling beginning in May 1963 and continuing through June 1964, the following may be concluded:

Small amounts of radioactivity are present in all sampled surface waters of Kentucky. Varying amounts of gross beta radioactivity were found in samples collected from all major streams during the period May 1963 to June

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1964. However, the concentration of radioactivity was well below the maximum permissible concentration limit for continuous exposure to the public as specified in the National Bureau of Standards Handbook 69 (3).

Radioactive waste from various industrial, educational, or medical isotope uses are not presently contributing noticeably to radioactivity levels in the streams.

The concentration of suspended radioactivity in surface waters is extremely variable. Suspended beta radioactivity concentrations in water samples collected throughout the State have ranged from less than 2 pc/liter to over 43 pc/liter. The dissolved beta radioactivity content of most streams appears relatively stable as compared to that in the suspended fraction.

Under normal conditions of streamflow the dissolved fraction accounts for essentially all the beta radioactivity. As flow increases the suspended radioactivity content increases for a short period. However, this relationship is not necessarily linear. This would appear to be related to an increase in total suspended solids resulting from increased turbulence.

REFERENCES

 U. S. Geological Survey. Surface Water Records of Kentucky. U. S. Department of the Interior, District Engineer, Surface Water Branch, U. S. Geological Survey, 522 West Jefferson Street, Louisville 2, Kentucky.

Division of Radiological Health. Radionuclide Analyses of Environmental Samples R59-6, Radiological Health Research Activities, Robert A. Taft Sanitary Engineering Center, Cincinnati 26, Ohio (November

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16, 1959).

3. National Bureau of Standards. Maximum Permissible Body Burdens and Maximum Permissible Concentration of Radionuclides in Air and in Water for Occupational Exposure. Handbook 69, U. S. Government Printing Office, Washington, D. C. price \$0.35.

Section IV—Other Data

RADIUM SOURCE SURVEYS IN KENTUCKY

Radiological Health Program
Kentucky Department of Health

Radium sources constitute a potential radiation hazard because of their ready availability. and because improper handling or storage may lead to loss or rupture (due to internal gas pressure). To protect the health of the citizens of Kentucky from these and other sources of ionizing radiation, regulations were adopted in October 1962 by the Kentucky State Department of Health requiring the registration of naturally-occurring or accelerator-produced radioactive materials and also radiation-producing machines. As a result of the implementation of these regulations, a total of 46 radium users were registered by December 1963. More than 80 percent of the sources registered were in the possession of hospitals or private physicians (see table 1).

TABLE 1.—NUMBERS OF RADIUM USERS

	Type of facility		No. of facilities
Hospital			2
Private phys	ician		10
linic	titution		
ndustrial			
Veterinarian			

At the time of source registrations, each user was provided with a copy of the radiation regulations and notified on his registration certificate of the need for annual leak tests of each radium source. They were also provided with a suggested leak-test procedure.

The 46 registered users of radium in the State were surveyed during December 1963 and January 1964. A typical survey provided information regarding owner, type of facility, typical users, and information regarding radiological health procedures practiced at the facility. Results of the Statewide radium survey indicated that concern for radiological health aspects of radium storage and handling was justified (see table 2). The percentages of noncompliance items indicated in table 2, although high, would have been much higher except that State Health Department staff took time to correct many items of noncompliance at the time of the survey.

TABLE 2.—RADIUM SURVEY RESULTS

Items needing correction	Percent of total surveyed
No records of any type of survey performed	5 4 3
Radium container not posted (warning signs) Areas not posted (warning signs) Personnel instruction or training not accomplished Radiation surveys of any type not performed	
Improper storage (unnecessary exposure) No routine operating procedures. No records of personnel exposures (film badges, etc.)	i

¹ Regulations require personnel monitoring where an individual is likely to receive in excess of 25 percent of the occupational exposure limit in any calendar quarter.

Discussions with the various users of radium indicated lack of respect for the significance of leaking sources of contamination therefrom. It appeared that most users were more concerned with the monetary value of the source rather than the potential radiation or contamination problem. Obviously, immediate action was necessary to check for possible leaking sources and contamination, and to provide some guidance and advice concerning proper storage and handling procedures. After this initial information was obtained, a survey kit was prepared consisting of an Eberline PAC-1SA alpha scintillation detector instrument, a Nuclear Chicago Model 2650 instrument with thin-wall and end-window G-M tubes, a box of paper tissue for obtaining smear samples, cotton-tipped sticks and alcohol to clean possibly contaminated detector probe faces, and a calibrated alpha check source of approximately 11,000 dpm (two-pi).

Direct measurements of radiation intensity and that of wipe samples from the sources were made. A wipe sample measuring greater than 200 cpm (with a 2-pi calibrated detector) of alpha activity or its equivalent 0.2 mrad/hr (by end window G-M) of alpha-beta-gamma activity was considered as the "alert level" for further leak testing. It was recommended that these further leak tests be performed on each source by health physics or radiological physics consultants. All collected wipe samples indicating potential contamination were returned to the laboratory and analyzed with internal gasflow proportional counters to verify results of portable instrument monitoring.

Effectiveness of the survey technique was verified by the consultants finding leaking sources each time radiocontamination was at or above the "alert level." No "leaks" were found in those areas where no radiocontaminathree typical surveys were as indicated in table tion was found above "alert levels." Results of 3.

Test results verified that leaking sources could be readily detected by the use of portable instruments in a program of meticulous surveys of the storage area and storage container, and measurements of wipe-test samples from suspect areas.

Survey report and results forms were prepared not only to assure an orderly and complete recording of data but primarily to provide an immediate report to the user for suggested corrective action. Items of information required on the form were established to assure compliance with the Kentucky Department of Health radiation regulations. At the conclusion of each facility survey, the user was given a result form with the various items of noncompliance indicated and an explanation of what was required to correct the deficiencies. A copy of this form was returned to the Radiological Health Program records after it had been signed by the facility Radiation Safety Officer, or other responsible individual, at the conclusion of the survey discussion.

Table 3.—MEASURED CONTAMINATION IN THREE TYPICAL SURVEYS

Sample	Health Department results	Consultant results
Smear inside containerCotton from 24-hour storageCotton from 24-hour storage	500 c/m (alpha)	3,000 c/m (alpha) b 0.4 mrad/hr (beta & gamma) with thin-wall G-M.
Smear inside container	No contamination	No contamination
Smear inside container Direct measurement on lid of container Cotton from 24-hour storage	CASE #3 1,000 c/m (alpha) 10,000 c/m (alpha)	• 100,000 c/m with well-type gamma scintillator.

Dash indicates no measurement done.
 The beta and gamma measurement would have been approximately 3 mrad/hr with an end-window G-M

The radioactivity on the sealed cotton from 24-hour closed test tube storage, after the source was rendecayed with a half-life of approximately 10 hours.

The importance of radiological health surveys of radium users became apparent when a radiation physicist called attention to a leaking radium source that his organization had been given-property from the estate of a deceased physician. The leaking source was properly stored and measurements were made to assure insignificant contamination of equipment and property of the deceased physician. The leaking source and three other radium needles stored in the same container were brought to the Health Department Radiological Laboratory for further study and evaluation of proposed survey techniques and leak test procedures. At the conclusion of these tests, the leaking source was disposed of in an approved manner.

The success of this program was primarily due to the cooperation given survey teams, and interest and corrective action taken by the users in all cases. It is felt that the radium survey, and continuing program of radium source annual surveys, is one of the milestones in the Kentucky Radiological Health Program.

TABLE 4.—PORTABLE INSTRUMENT MEASUREMENTS

III Too and the	Cot	Smear	
Needle No.	Beta+gamma 1 (mrad/hour)	Alpha+beta 2 (mrad/hour)	Alpha 3 (cpm)
1 2 3 4	4 BG BG 10 BG	BG BG 300 (off scale) 2.5	BG BG 20,000

Determined by Geiger-Mueller detector.
 Determined by low-energy ion chamber.
 Determined by gas-flow internal proportional counter.
 BG indicates reading cannot be distinguished from background.

TABLE 5.—GAMMA SPECTRUM RESULTS

Gamma peak (MEV)	Relative peak activity				
111111111111111111111111111111111111111	Needle	Wash water	Cotton		
0.10	7.0	7.5	7.0		
0.18	5.0	6.0	2.4		
0.30	4.0	3.8	4.1		
0.35	4.0	4.0	5.4		
0.61	2.3	2.2	3.0		
1.0	0.4	0.5	0.4		
1.1	0.6	0.6	0.3		
1.43	0.5	0.4	0.4		
1.75	0.3	0.4	0.4		

ENVIRONMENTAL LEVELS OF RADIOACTIVITY AT ATOMIC ENERGY COMMISSION INSTALLATIONS

The U.S. Atomic Energy Commission receives from its contractors periodic reports on the environmental levels of radioactivity in the vicinity of major Commission installations. The reports include data from routine monitoring programs where operations are of such a nature that plant perimeter surveys are required.

Summaries of the environmental radioactivity for 23 AEC installations have appeared periodically in RHD since November 1960. Following are reports for Argonne National Laboratory, Atomics International, and National Reactor Testing Station.

Releases of radioactive materials from these installations for the periods covered in the reports below are governed by standards set forth in appropriate chapters of the AEC manual. The radioactivity concentration limits applicable to effluents released from AEC installations are essentially identical to those published in the Federal Register (1).

1. Argonne National Laboratory, January-June 1964

University of Chicago Lemont, Illinois

Air monitoring

Weekly continuous samples of airborne particulates for radioactivity measurements at 7 locations on the Argonne National Laboratory (ANL) site as shown in figure 1 and at 5 offsite locations at Aurora (west of ANL site). Wheaton (northwest), Hinsdale (northeast), Joliet (southwest), and Tinley Park (southwest). The quarterly averages of alpha, beta, and several nuclide concentrations are given in table 1. The radionuclide determinations were made from gamma spectra of monthly composites of the filters. The data show little difference between offsite and onsite measurements for alpha activity and most of the nuclides, in-

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TABLE 1.—RADIOACTIVITY OF AIRBORNE PARTICULATES, ANL, JANUARY-JUNE 1964

[Average concentrations in pc/m³]

Type of analysis	Sampling locations	January	February	March	April	May	June	Average
Alpha	On site	0.0032 0.0029	0.0036 0.0033	0.0039 0.0042	0.0042 0.0042	0.0051 0.0043	0.0039 0.0048	0.0040
Beta	On site	0.9	1.0	1.4	1.5	2.0 2.1	1.8	1.4
Sb125	On siteOff-site	0.02 0.02	0.02 0.02	0.04 0.04	0.05 0.04	0.06	0.06	0.04 0.04
Ce144	On siteOff-site	0.54 0.53	0.52 0.48	0.63 0.72	0.80 0.69	1.10	0.85 1.06	0.74 0.75
Cs137	On siteOff-site	0.04 0.04	0.04 0.04	0.07 0.07	0.07 0.08	0.11 0.11	0.11 0.10	0.07 0.07
Ru ¹⁶³	On site	<0.05 <0.05	<0.05 <0.05	0.06 0.06	0.05	0.06	0.04 0.03	0.04 0.04
Ru-Rh ¹⁹⁶	On siteOff-site	0.24 0.24	0.21 0.24	0.27 0.28	0.28 0.33	0.39 0.32	0.38 0.43	0.30 0.31
Zr-Nb ⁹⁵	On site	0.09	0.07	0.09	0.08	0.09	0.07 0.07	0.08

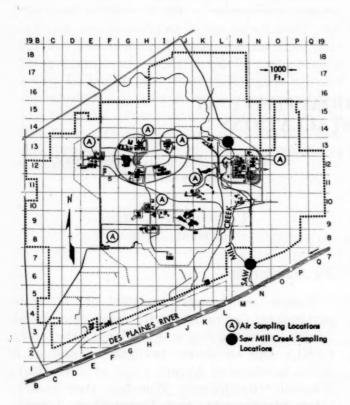


FIGURE 1.—ONSITE SAMPLING LOCATIONS, ARGONNE NATIONAL LABORATORY

dicating that ANL does not contribute detectable quantities of these activities to the atmosphere. The average airborne beta activity during the first half of 1964 was approximately 1.5 pc/m³.

Water monitoring

ANL waste water is discharged into Sawmill Creek, a stream that runs through the Argonne Laboratory site and enters the Des Plaines River about 500 yards downstream from the waste water discharge. Sampling locations on Sawmill Creek and Des Plaines River are shown in figures 1 and 2, respectively.

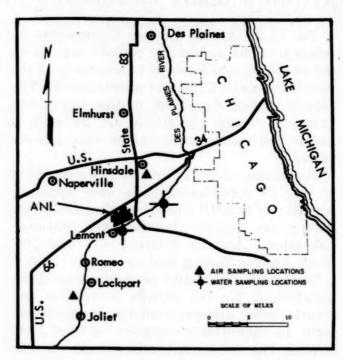


FIGURE 2.—SITE LOCATION OF ARGONNE NATIONAL LABORATORY (INCLUDING SOME OFFSITE SAMPLING STATIONS)

TABLE 2.—RADIOACTIVITY IN SAWMILL CREEK, ANL, JANUARY-JUNE 1964

[Average concentrations in pc/liter]

Type of analysis	Sampling location	January-June 1964		
1/ 1/11		Maximum	Minimum	Average
Alpha emitters: Total alpha	Upstream Downstream	3.8	0.6	1.5
U-natural	Upstream Downstream	2.5 8.3	0.3	1.5 3.0
Pu ²³³	Upstream	<0.05	<0.05	<0.08
	Downstream	0.35	<0.05	0.10
Th ²³²	Upstream	<0.05	<0.05	<0.00
	Downstream	0.60	<0.05	0.13
Beta emitters:	Upstream	73	7	25
Total beta	Downstream	42	13	20
Coss	Upstream	<5	<5	<5
	Downstream	62	<5	5
Co ⁶⁰	Upstream	<3	<3	<3
	Downstream	40	<2	3
Sr**	Upstream	<2	<2	<2
	Downstream	<2	<2	<2
Sr ⁹⁰	Upstream Downstream	4.5	<0.5 <0.5	3.4 1.8
I181	Upstream	<3	<3	<3
	Downstream	<3	<3	<3
Ce137	Upstream Downstream	0.8	<0.5 <0.5	0.4
Ba ¹⁴⁰	Upstream	<2	<2	<2
	Downstream	<2	<2	<2
Th-Pa ²³⁴	Upstream Downstream		0.6	1.2

On Sawmill Creek, weekly grab samples are collected upstream and 3 times a week samples are collected downstream from the waste water outfall. The upstream flow is roughly equal to the waste water flow, yielding a dilution factor of one-half. The data in table 2 show higher concentrations downstream and upstream. However, downstream concentrations were well below permissible concentration limits.

Weekly grab samples are collected from the Des Plaines River upstream and downstream from its junction with Sawmill Creek. The results indicate that the dilution factor of the Des Plaines River is so large that the radioactivity contribution from ANL is not detectable.

Recent coverage in Radiological Health Data:

Period	Issue
Third and fourth quarters 1960	July 1961
First and second quarters 1961	December 1961
Third and fourth quarters 1961	May 1962
1962	May 1963
1963	August 1964

REFERENCE

U.S. Atomic Energy Commission. Rules and regulations, standards for protection against radiation.
 Federal Register. Title 10, Part 20, Appendix B,
 Table 2. Superintendent of Documents, U.S. Government Printing Office, Washington, D.C. 20402.

2. Atomics International January-June 1964

Canoga Park, California

Atomics International, a division of North American Aviation, Incorporated, operates the Nuclear Development Field Laboratory (NDFL) and the World Headquarters Facility (WHF) at Canoga Park, California, under contract with the Atomic Energy Commission. The company designs, develops, and constructs nuclear reactors for control stations, compact power plants, and medical, industrial, and scientific applications. Locations of the facilities are shown in figure 3.

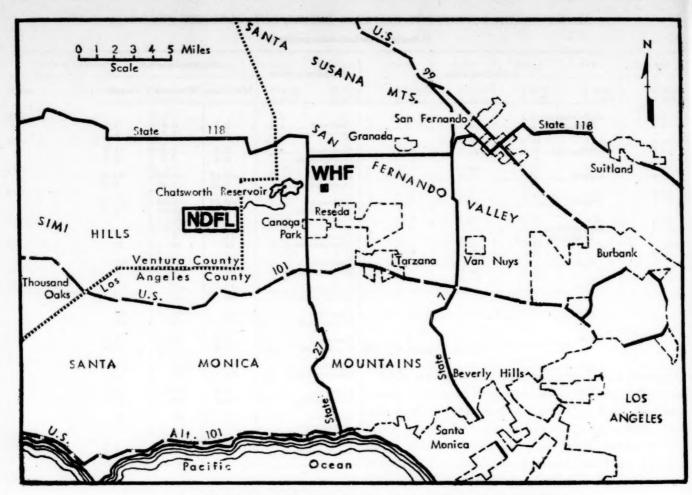


FIGURE 3.—ATOMICS INTERNATIONAL FACILITIES AND VICINITY

The NDFL facilities include a 20-megawatt sodium reactor experiment (SRE) power reactor; several smaller experimental reactor facilities such as critical facilities, Systems for Nuclear Auxiliary Power (SNAP) reactor, shield test facilities, and others; and extensive rolling and fuel fabrication operations. The WHF is primarily an administrative building, but a small amount of fuel fabrication is conducted there. For this reason, the WHF area is included in the environmental monitoring.

Environmental monitoring is conducted at WHF and NDFL to test the effectiveness of radiological safety procedures and of engineering safeguards incorporated into facility design. Surface soil, vegetation, water and air samples are surveyed periodically.

Air monitoring

Environmental air sampling is conducted continuously at the WHF and NDFL sites with automatic air samplers operating on 24-hour sampling cycles. Airborne particulate radioactivity is collected on a stationary filter tape which is automatically changed at the end of each sampling period. The filter tape is removed, allowed to decay for 72 hours, and counted in an automatic proportional counting system. The volume of atypical daily air sample is approximately 21 cubic meters. Averages for beta-gamma activity in airborne particulates are given in table 3.

Water samples

Processed water at the NDFL is obtained from wells and stored in 50,000 gallon tanks. Monthly samples are collected from these wells (see table 4). Potable water is obtained in bottles delivered to the site and is, therefore, not analyzed. Process water is drawn from the wells into one-liter polyethylene bottles for transfer to the laboratory. Water samples from the lake surface and supply inlet of Chatsworth

TABLE 3.—AIRBORNE PARTICULATE RADIOACTIVITY DATA

[activity in pc/m3]

	19	963	First half 1964		
Site	Number of samples	Average concentration	Number of samples	Average concentration	
HeadquartersNDFL	360 292	6.6	182 69	2.0	

TABLE 4.—WELL WATER MONITORING, NDFL

[pc/liter]

	. 196	3	First half 1964	
Activity	Number of samples	Average	Number of samples	Average
α β, γ	24 24	*0.18 7.0	11 11	*0.12 *4.7

[·] Rounded from range of values originally reported.

Reservoir are similarly obtained (see table 5). In some cases the average concentration is presented as a range. The range occurs when one or more of the samples analyzed contains an undetectable amount of radioactivity.

TABLE 5.—CHATSWORTH RESERVOIR WATER SAMPLING

[pc/liter]

		19	63	First half 1964	
Sample	Activity	Number of samples	Average	Number of samples	Average
Lake surface	β, γ	37 37	0.84 18	12 12	0.60 12.6
Supply inlet	β, γ	12 12	*0.58	6	*0.41 10.

^{*} Rounded from range of values originally reported.

In the laboratory, 500 ml. of water are evaporated to dryness in crystallizing dishes at approximately 90°C. The residue salts are transferred into stainless steel planchets, wetted to produce an even sample distribution, redried under infra-red lamps, and counted in an automatic proportional counting system.

Soil and vegetation samples

Soil and vegetation are sampled monthly at 51 locations. Thirteen sampling stations are located within the boundaries of Atomics International and are designated as on site stations. The remaining 38 stations are located within a 10-mile radius of Atomics International and are referred to as off-site stations. Several of the off-site stations are located at Chatsworth Reservoir, which is operated by the Los Angeles City Department of Water and Power.

Surface soil samples range from decomposed granite to clay and loam and are collected from the top half-inch layer of ground surface. The samples are placed in plastic containers and sent to the laboratory. Sample preparation consists of ashing in a muffle furnace at 500°C. for approximately eight hours, cooling, and sieving to obtain uniform particles for counting. One-gram aliquots of the soil are put in stainless steel planchets, wetted with acetone, redried, and counted in an automatic proportional counting system. Alpha and beta-gamma activities of soil samples are given in table 6.

TABLE 6.—SOIL RADIOACTIVITY

[pc/g]

	196	63	First half 1964	
Activity .	Number of samples	Average	Number of samples	Average
αβ-γ	156 156	* 0.42 45	74 74	* 0.35
β-γ	455 455	* 0.40 42	221 221	* 0.32
	α	Activity Number of samples α	Number of samples Average α	Activity Number of samples Average of samples Number of samples α

^{*} Rounded from range of values originally reported.

Vegetation samples obtained in the field at each station are generally sunflower or wild tobacco plant leaves. At the laboratory the leaves are stripped from the plant and washed to remove foreign matter. The vegetation, after a distilled water rinse, is ashed in a muffle furnace at 500°C. for approximately eight hours, producing a completely oxidized ash of uniform density. Three-hundred milligram aliquots of ground ash are used for counting in an automatic proportional counting system. Vegetation data are shown in table 7.

TABLE 7.—RADIOACTIVITY IN VEGETATION

		196	63	First half 1964				
Area	Activity	Number of samples	Average	Number of samples	Average			
On-site	α β-γ	156 156	* 0.44 465	76 76	* 0.39 247			
Off-site	β-γ	456 456	*0.37 388	227 227	* 0.39 231			

a Rounded from range of values originally reported.

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Previous coverage in Radiological Health Data:

Period
1960 and First and
Second quarters 1961
Third and Fourth quarters 1961
Calendar Year 1962
Calendar Year 1963
Calendar Year 1963

December 1961
August 1962
November 1963
September 1964

3. National Reactor Testing Station January-June 1964

Health and Safety Division Atomic Energy Commission Idaho Falls, Idaho

The National Reactor Testing Station (NRTS) is located about 20 miles from Idaho Falls, Idaho, in a very remote area which in a large measure permits controlled releases of radioactivity from the projects with minimum risk to the environs.

Responsibility for holding the environmental radioactivity levels at the station below the Radiation Protection Guides (RPG) recommended by the Federal Radiation Council (FRC) lies with the Atomic Energy Commission. One of the safeguards employed is a monitoring program. Descriptions of the monitoring network procedures have been presented earlier (see previous coverage). Beginning in January 1964, air data was reported by a radiation telemitry system while the former dual filter system continued in operation to obtain comparative data. The new system includes data from ion chambers and results of G-M counts from filter paper tapes. Quarterly environmental monitoring results in addition to averages for January to June 1964, are shown in table 8. Sampling locations at NRTS can be seen in figure 4.

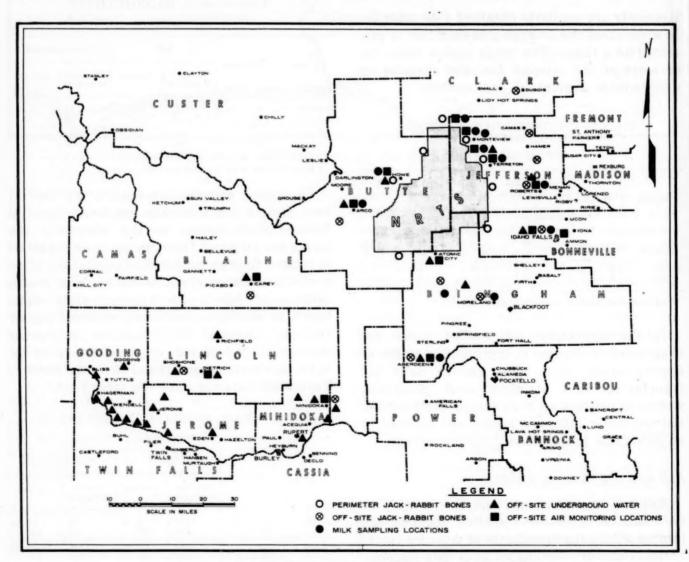


FIGURE 4.—ENVIRONMENTAL MONITORING STATIONS

TABLE 8.—ENVIRONMENTAL MONITORING PROGRAM DATA, NRTS, JANUARY-JUNE 1964

Type of sample and units	Number of stations	Approximate frequency of collection	Type of analysis	Minimum level of detection	Maximum activity of single sample	Average activity per sample
Off-site underground water (pc/liter)	32	semi-annual	alphabeta	3 6	9 20	<4 <10
On-site production well water (pc/liter)	22	two weeks	alphabeta	3	7	<4
Off-site air (pc/m³)	15 15	continuous	beta-gammaiodine-131	1.6 3.6	100 19 40	<1.6 <20 <1.6 <3.6
Off-site milk (pc/liter)	14 11	monthly	iodine-131strontium-90	20 1.5	<20 44	<20 28
Off-site area monitoring badges (mrem)	14	monthly	betagamma	10 10	10 25	<60 <60

Previous coverage in Radiological Health Data:

Period	
1959 ar	d first quarter 1960
Second	quarter 1960
	and fourth quarters 1960
	nd second quarters 1961
Third :	and fourth quarters 1961
1962	
1963	

Issue
November 1960
February 1961
May 1961
January 1962
June 1962
June 1963
September 1964

REPORTED NUCLEAR DETONATIONS, FEBRUARY 1965

During the month of February three United States nuclear tests were announced by the Atomic Energy Commission.

The February 4, 16, and 18 tests were conducted underground at the Nevada Test Site. All were of low-yield (less than 20 kilotons).